

**Introduction:**

When you bake/cook something, we use a specific amount of each ingredient. Imagine if you made a batch of cookies and used way too many eggs, or not enough sugar. YUCK! In chemistry, reactions proceed with very specific recipes. The study of these recipes is stoichiometry. When the reactants are present in the correct amounts, the reaction makes products. We are going to look at what happens when there are more or less of some of the reactants present.

**Procedure:** PHET Simulations → Play with the Sims → Chemistry → Reactions, Products, and Leftovers OPEN IN HTML 5  
[http://phet.colorado.edu/en/simulations/available-for-education-products-and-leftovers/html5-reactions-products-and-leftovers\\_molecule](http://phet.colorado.edu/en/simulations/available-for-education-products-and-leftovers/html5-reactions-products-and-leftovers_molecule)

**Part 1: Making Sandwiches**



1. A cheese sandwich requires 2 slices of bread to 1 slice of cheese. If there is exactly twice as many slices of bread as cheese, then there will be no leftovers. What happens if you have two slices of bread and one slice of cheese? You have a slice of cheese left over!
2. Use the simulation to play around with this two-reactant synthesis reaction. Take some time and familiarize yourself with the simulation buttons.
3. At the bottom of the screen, change the coefficients of the reactants to 2:1.



4. Notice that there is now one sandwich with no leftovers!
5. Complete situations B through D. A has been done for you as an example.

| Situation | Bread Used | Cheese Used | Sandwiches Made | Excess Bread | Excess Cheese |
|-----------|------------|-------------|-----------------|--------------|---------------|
| A         | 4 slices   | 2 slices    | 2 sandwiches    | 0 bread      | 0 cheese      |
| B         | 2 slices   | 3 slices    |                 |              |               |
| C         |            |             | 2 sandwiches    | 1 slice      | 0 slices      |
| D         | 6 slices   |             | 3 sandwiches    |              | 4 slices      |

6. Look at situation A. The cheese in this case is said to be in excess. The other reactant (bread) has no leftovers. The bread in this case is the limiting reactant.
7. In situations B, C, and D, determine the limiting reactant: B: \_\_\_\_\_ C: \_\_\_\_\_ D: \_\_\_\_\_



**Part 2: Real Chemical Reactions**



8. Select "Molecules" towards the bottom of your screen. We can now work with real chemical reactions, one that creates a very important molecule!
9. What is the lowest molecular ratio for the reaction of hydrogen and oxygen to produce water? That is, balance the following equation:



10. Complete the table below while making water H<sub>2</sub>O from hydrogen H<sub>2</sub> and oxygen O<sub>2</sub>.

| Hydrogen Molecules H <sub>2</sub> | Oxygen Molecules O <sub>2</sub> | Water Molecules H <sub>2</sub> O | Excess H <sub>2</sub> | Excess O <sub>2</sub> |
|-----------------------------------|---------------------------------|----------------------------------|-----------------------|-----------------------|
| 4 molecules                       | 4 molecules                     |                                  |                       |                       |
| 7 molecules                       | 6 molecules                     |                                  |                       |                       |
|                                   |                                 | 4 molecules                      | 0 molecules           | 0 molecules           |
| 6 molecules                       | 8 molecules                     |                                  |                       |                       |
|                                   |                                 | 4 molecules                      | 1 molecules           | 0 molecules           |

Stop for check point. Teacher signature for this page: \_\_\_\_\_

# Basic Stoichiometry Phet Lab Homework Exercises Answers

**Michael Seilmaier**



**Basic Stoichiometry Phet Lab Homework Exercises Answers:**

The book delves into Basic Stoichiometry Phet Lab Homework Exercises Answers. Basic Stoichiometry Phet Lab Homework Exercises Answers is an essential topic that needs to be grasped by everyone, from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Basic Stoichiometry Phet Lab Homework Exercises Answers, encompassing both the fundamentals and more intricate discussions.

1. This book is structured into several chapters, namely:
    - Chapter 1: Introduction to Basic Stoichiometry Phet Lab Homework Exercises Answers
    - Chapter 2: Essential Elements of Basic Stoichiometry Phet Lab Homework Exercises Answers
    - Chapter 3: Basic Stoichiometry Phet Lab Homework Exercises Answers in Everyday Life
    - Chapter 4: Basic Stoichiometry Phet Lab Homework Exercises Answers in Specific Contexts
    - Chapter 5: Conclusion
  2. In chapter 1, this book will provide an overview of Basic Stoichiometry Phet Lab Homework Exercises Answers. The first chapter will explore what Basic Stoichiometry Phet Lab Homework Exercises Answers is, why Basic Stoichiometry Phet Lab Homework Exercises Answers is vital, and how to effectively learn about Basic Stoichiometry Phet Lab Homework Exercises Answers.
  3. In chapter 2, the author will delve into the foundational concepts of Basic Stoichiometry Phet Lab Homework Exercises Answers. This chapter will elucidate the essential principles that must be understood to grasp Basic Stoichiometry Phet Lab Homework Exercises Answers in its entirety.
  4. In chapter 3, this book will examine the practical applications of Basic Stoichiometry Phet Lab Homework Exercises Answers in daily life. This chapter will showcase real-world examples of how Basic Stoichiometry Phet Lab Homework Exercises Answers can be effectively utilized in everyday scenarios.
  5. In chapter 4, the author will scrutinize the relevance of Basic Stoichiometry Phet Lab Homework Exercises Answers in specific contexts. The fourth chapter will explore how Basic Stoichiometry Phet Lab Homework Exercises Answers is applied in specialized fields, such as education, business, and technology.
  6. In chapter 5, this book will draw a conclusion about Basic Stoichiometry Phet Lab Homework Exercises Answers. The final chapter will summarize the key points that have been discussed throughout the book.
- This book is crafted in an easy-to-understand language and is complemented by engaging illustrations. This book is highly recommended for anyone seeking to gain a comprehensive understanding of Basic Stoichiometry Phet Lab Homework Exercises Answers.

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