

4. pH of Prepared Buffer Solutions

Buffer Solution	Acetic Acid; 4.5%, mL	Sodium Acetate Solution, mL	Recorded pH
A	5 mL	5 mL	4.7
B	5 mL	1 mL	3.9
C	10 mL	1 mL	3.7
D	1 mL	10 mL	5.7
E	1 mL	5 mL	5.3

III. Calculations/Results

*show all calculations with units

1. Sodium acetate moles _____
(molar mass = 82.0 g/mol)
2. Molarity of sodium acetate = _____
3. Concentration of acetic acid (from bottle) = _____
4. Molarity of acetic acid = _____
(molar mass = 60.0 g/mol; density of the solution is 1.0 g/mL)

Buffer Solution Lab Report

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Laboratory Methods in Dynamic Electroanalysis M. Teresa Fernández Abedul, 2019-10-13 *Laboratory Methods in Dynamic Electroanalysis* is a useful guide to introduce analytical chemists and scientists of related disciplines to the world of dynamic electroanalysis using simple and low cost methods. The trend toward decentralization of analysis has made this fascinating field one of the fastest growing branches of analytical chemistry. As electroanalytical devices have moved from conventional electrochemical cells (10-20 mL) to current cells (e.g. 5-50 mL) based on different materials such as paper or polymers that integrate thick or thin film electrodes, interesting strategies have emerged such as the combination of microfluidic cells and biosensing or nanostructuring of electrodes. This book provides detailed easy procedures for dynamic electroanalysis and covers the main trends in electrochemical cells and electrodes including microfluidic electrodes, electrochemical detection in microchip electrophoresis, nanostructuring of electrodes, development of bio enzymatic immuno and DNA assays, paper based electrodes, interdigitated array electrodes, multiplexed analysis and combination with optics. Different strategies and techniques: amperometric, voltammetric and impedimetric are presented in a didactic practice based way and a bibliography provides readers with additional sources of information. Provides easy to implement experiments using low cost simple equipment. Includes laboratory methodologies that utilize both conventional designs and the latest trends in dynamic electroanalysis. Goes beyond the fundamentals covered in other books focusing instead on practical applications of electroanalysis.

Chemistry in the Laboratory James M. Postma, Julian L. Robert, J. Leland Hollenberg, 2004-03-12 This clearly written class tested manual has long given students hands on experience covering all the essential topics in general chemistry. Stand alone experiments provide all the background introduction necessary to work with any general chemistry text. This revised edition offers new experiments and expanded information on applications to real world situations.

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