

Calculations with Moles: Using Standard Molar Volume

How many liters are occupied by 3 moles of oxygen gas at STP?

$$\frac{3 \text{ mol}}{1 \text{ mol}} \times \frac{22.4 \text{ L}}{1 \text{ mol}} = 67.2 \text{ L}$$

Calculations Using Standard Molar Volume

Clifford Lane



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