

## WSET Level 3 Systematic Approach to Tasting Spirits®

APPEARANCE		
Clarity		clear – hazy (faulty?)
Intensity		water-white – pale – medium – deep – opaque
Colour		colourless – lemon – gold – amber – brown pink – red – orange – yellow – green – blue – purple – brown – black
Other observations	louching other	slightly hazy – cloudy and opaque particles
NOSE		
Condition		clean – unclean (faulty?)
Aroma Intensity		neutral – light – medium(-) – medium – medium(+) – pronounced
Aroma characteristics		e.g. raw materials, processing, oak and maturation
PALATE		
Sweetness		dry – off-dry – medium – sweet
Flavour intensity		neutral – light – medium(-) – medium – medium(+) – pronounced
Texture/other		e.g. rough, smooth, watery, mouthfilling, warming e.g. bitterness
Flavour characteristics		e.g. raw materials, processing, oak and maturation
Finish	length nature	short – medium(-) – medium – medium(+) – long neutral – simple – some complexity – very complex
CONCLUSIONS ASSESSMENT OF QUALITY		
Building an argument		e.g. balance, length and intensity, complexity, expressiveness
Quality level		faulty – poor – acceptable – good – very good – outstanding

### Notes to students:

For lines where the entries are separated by a hyphen – You must select one and only one of these options.

For lines starting with "e.g." where the entries are separated with commas – These are entries that you should consider when writing your tasting note. You may not need to comment on each entry for every spirit.



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# Wset Level 3 Systematic Approach To Tasting Wine Sat

**KJ Lindholm-Leary**



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web sec 16.5 page 743 solution sec 12.3.12.4 page 519 in a saturated potassium nitrate  $\text{KNO}_3$  solution in water h<sub>2</sub>o a dynamic equilibrium will be established and the

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web the amount of  $\text{KNO}_3$  in the solution is called solubility if the  $\text{KNO}_3$  is less than what the water can handle then it is an unsaturated solution if the amount of  $\text{KNO}_3$  is equal to

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web nov 4 2016 potassium nitrate is exceptionally soluble in water your solution is unsaturated a solution is saturated if it contains the same amount of solute as

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web may 2 2021 an aqueous  $\text{KNO}_3$  solution has a molality of 4.16 m and a density of 1.08 g/ml calculate the percentage by mass  $\text{KNO}_3$

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potassium nitrate has an orthorhombic crystal structure at room temperature which transforms to a trigonal system at 128 °C (262 °F) on cooling from 200 °C (392 °F) another trigonal phase forms between 124 °C (255 °F) and 100 °C (212 °F) sodium nitrate is isomorphous with calcite the most stable form of calcium carbonate whereas room temperature potassium nitrate is isomorphous with aragonite a slightly less stable polymorph of

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