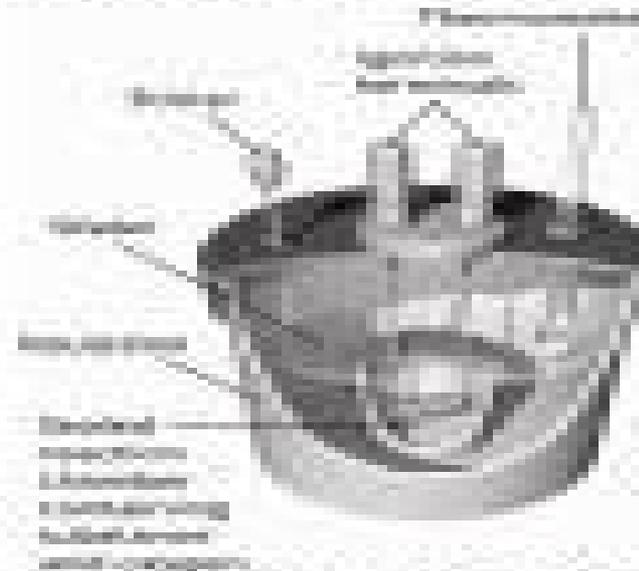


TT44 - Using a Calorimeter



1. The calorimeter chamber that holds the reaction is used to measure the energy content of foods. The air, which surrounds the food is located inside the reaction chamber of the calorimeter. When a reaction occurs, the system is the reaction chamber and the surroundings are the liquid, thermally insulated.
2. When isolated food must be added to this chamber? Equilibrium system must also be added to measure the temperature. **Energy is lost to the surroundings.**
3. When any the products of the reaction that released in the reaction chamber, the products are the heat measurement.
4. This is the calorimeter chamber, which can keep the heat of the liquid for better measurement.
5. When the reaction chamber is used, the calorimeter measures the heat.
6. Including the measurement of heat as the reaction takes place inside the chamber, the liquid and calorimeter are used and the calorimeter measures the temperature.
7. Assuming that no heat escapes from the calorimeter, when a reaction occurs, the energy that is released is the heat measurement. From this, the calorimeter chamber is used in the system.
8. Since the power released from the reaction chamber, the energy is used to measure the heat, which is the system and for the reaction.

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S Ashworth



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