

# Using Mole Ratios

Use with Chapter 11,  
Section 11.2

Mole ratios relate moles of unknown and known substances in a balanced chemical equation and are used to make stoichiometric mole-to-mole conversions between molar amounts of unknown and known substances.



To convert

number of moles of known substance	to	number of moles of unknown substance	use this	mole ratio
$\text{C}_3\text{H}_8$		$\text{O}_2$		_____
$\text{C}_3\text{H}_8$		$\text{CO}_2$		_____
$\text{C}_3\text{H}_8$		$\text{H}_2\text{O}$		_____
$\text{O}_2$		$\text{C}_3\text{H}_8$		_____
$\text{O}_2$		$\text{CO}_2$		_____
$\text{O}_2$		$\text{H}_2\text{O}$		_____
$\text{CO}_2$		$\text{C}_3\text{H}_8$		_____
$\text{CO}_2$		$\text{O}_2$		_____
$\text{CO}_2$		$\text{H}_2\text{O}$		_____
$\text{H}_2\text{O}$		$\text{C}_3\text{H}_8$		_____
$\text{H}_2\text{O}$		$\text{O}_2$		_____
$\text{H}_2\text{O}$		$\text{CO}_2$		_____

# Teaching Transparency Master 39 Chemistry Answers

**Marco Cascella**



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