

5-2 Review and Reinforcement Part II

Quantum Theory and the Atom

Explain this principle: state the evidence for the discreteness of light (quantization).

photon(s)

Planck constant (symbol)

quantum mechanical model

quantum number(s)

electron density

orbitals

principal energy levels

_____ orbital _____ 1-4: model to explain around the nucleus where electrons in atoms really found.

_____ electron density _____ 2: the density of an electron cloud

_____ quantum number _____ 3: number describing a probability density found in atoms

Explain this principle: is there any experimental observation that supports quantum theory and this theory

quantum mechanical model explains photo

_____ quantum mechanical model _____ 2: experimentally predicted electron by knowing the electron's position and

momentum in atoms

_____ probability density _____ 3: the model's energy levels distribution

_____ electron spin _____ 4: the introduction of quantum mechanics model of atoms

Explain this principle: state the laws of the quantum mechanical model's construction

_____ 1. _____ 2: the electron cloud is everywhere where the probability of finding an electron is

- positive
- negative
- higher density
- continuous

_____ 4. _____ 3: the lowest principal energy level of the hydrogen atom contains only one

- orbital
- or orbitals
- of orbitals
- orbitals

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R Barnett



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