

Lab 5: Molecular Models



Table 1: Molecular Models

Example Atom (central atom is underlined>)	Number of Bonds on Central Atom (multiple bonds count as 1)	Unshared electron pairs on central atom	Regions of Electron Density	Molecular Geometry	Structure
<u>C</u> O ₂	2	0	2	Linear	
<u>Al</u> Cl ₃	3	0	3	Trigonal Planar	
<u>O</u> H ₂	2	2	4	Bent	
<u>N</u> H ₃	3	1	4	Trigonal pyramidal	
<u>C</u> Cl ₄	4	0	4	Tetrahedral	
<u>P</u> Cl ₅	5	0	5	Trigonal bipyramidal	
<u>S</u> F ₆	6	0	6	Octahedral	

Molecular Models Shapes Lab Answers

Kenda Jo Lemont



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