

# NACE CIP 2 EXAM QUESTIONS AND ANSWERS 100% PASS

Describe passivation - ✓✓ Passivation is the formation of a protective oxide film on the surface reducing its chemical activity and its ability to corrode

Describe the following factors and how they affect corrosion - ✓✓ Oxygen: oxygen increases the rate of corrosion temperature

Temperature: corrosion usually accelerated with increasing temperature

Chemical salts: increase the rate of corrosion by increasing the efficiency of the electrolyte

humidity: the wetter the environment, the more corrosion is likely to occur pollutants and acid gases: all promote corrosion

Two broad categories of corrosion can be described as - ✓✓ General and localized

Describe galvanic corrosion - ✓✓ Galvanic corrosion is an electrochemical action of two dissimilar metals in the presence of an electrolyte and an electron conductive path, which occurs when the dissimilar metals come into contact

# Nace Cip 2 Exam Questions

**Rachel Sandford**



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