

GRAPHING TRENDS OF THE PERIODIC TABLE

CHEMISTRY

NAME: _____

DATE: _____

PERIOD: _____ Graphing Periodic Table Trends

Directions: Using the provided graph paper, graph each of the elemental properties using a line graph. Place the property on the Y-axis and the Atomic number on the X-axis. Draw a dark, solid, vertical line on each graph where each row on the periodic table ends. (i.e. after atomic number 20). Use the graphs to analyze the trends in these properties and to answer the questions on the back.

Atomic Radius (pm)							
Lithium - 3	157	Sodium - 11	190	Potassium - 19	243	Rubidium - 27	295
Beryllium - 4	112	Magnesium - 12	145	Calcium - 20	194	Strontium - 28	219
Boron - 5	87	Aluminum - 13	175	Gallium - 31	136	Indium - 40	156
Carbon - 6	67	Silicon - 14	111	Germanium - 32	125	Tin - 50	145
Nitrogen - 7	56	Phosphorus - 15	98	Arsenic - 33	114	Antimony - 51	133
Oxygen - 8	48	Sulfur - 16	88	Selenium - 34	103	Tellurium - 52	123
Fluorine - 9	42	Chlorine - 17	79	Bromine - 35	94	Iodine - 53	119

Ionization Energy (kJ/mol)							
Lithium - 3	520	Sodium - 11	496	Potassium - 19	419	Rubidium - 27	403
Beryllium - 4	900	Magnesium - 12	738	Calcium - 20	590	Strontium - 28	550

Atomic Radius



Periodic Table Trends Assignment Answers

G Orfield



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Table of Contents Periodic Table Trends Assignment Answers

1. Understanding the eBook Periodic Table Trends Assignment Answers
 - The Rise of Digital Reading Periodic Table Trends Assignment Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Periodic Table Trends Assignment Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Periodic Table Trends Assignment Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Periodic Table Trends Assignment Answers
 - Personalized Recommendations

- Periodic Table Trends Assignment Answers User Reviews and Ratings
- Periodic Table Trends Assignment Answers and Bestseller Lists
- 5. Accessing Periodic Table Trends Assignment Answers Free and Paid eBooks
 - Periodic Table Trends Assignment Answers Public Domain eBooks
 - Periodic Table Trends Assignment Answers eBook Subscription Services
 - Periodic Table Trends Assignment Answers Budget-Friendly Options
- 6. Navigating Periodic Table Trends Assignment Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Periodic Table Trends Assignment Answers Compatibility with Devices
 - Periodic Table Trends Assignment Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Periodic Table Trends Assignment Answers
 - Highlighting and Note-Taking Periodic Table Trends Assignment Answers
 - Interactive Elements Periodic Table Trends Assignment Answers
- 8. Staying Engaged with Periodic Table Trends Assignment Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Periodic Table Trends Assignment Answers
- 9. Balancing eBooks and Physical Books Periodic Table Trends Assignment Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Periodic Table Trends Assignment Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Periodic Table Trends Assignment Answers
 - Setting Reading Goals Periodic Table Trends Assignment Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Periodic Table Trends Assignment Answers
 - Fact-Checking eBook Content of Periodic Table Trends Assignment Answers

- Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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