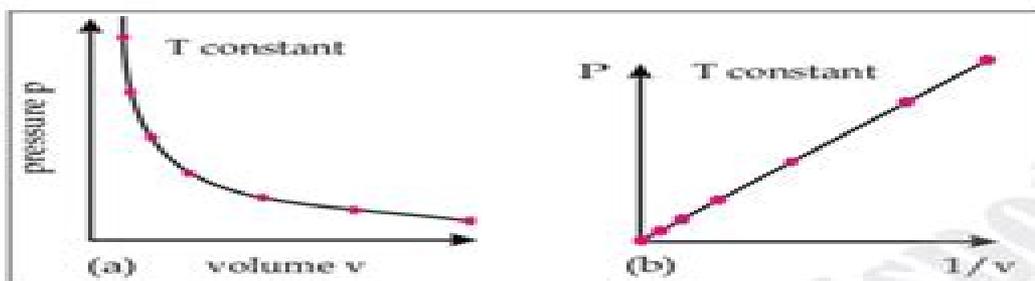


- (1) If the pressure of the gas is double volume becomes half. If pressure increases by three times then the volume becomes one-third and so on.
- (2) The graph between "p" and "V" between "p and  $\frac{1}{V}$ " is shown below:



- (3) The graph between "p" and "V" shows that if pressure increases then volume decreases and vice-versa.
- (4) The graph between "p" and " $1/V$ " shows a straight line passing through the origin.
- (5) At constant temperature the product of pressure and volume is constant. i.e.  
 $pV = \text{constant}$
- (6) Using the above result, at a constant temperature, we can write:  
 initial pressure  $\times$  initial volume = final pressure  $\times$  final volume  
 $p_1V_1 = p_2V_2$

Thus Robert Boyle concluded his Law known as "Boyle's Law" which states that:

**Statement of Boyle's Law:**

The volume of a fixed mass of a gas is inversely proportional to its pressure, provided its temperature remains constant.

**Q.23:** Write down the application of Boyle's Law.

**Ans:** Applications of (p-V) relationship of gas / Boyle's Law:

Some applications of pressure-volume (p-V) relationship of a gas i.e. Boyle's Law is given below:

Aerosols, such as spray paints, use Boyle's Law in their working mechanism.

When we breathe, our diaphragm moves downward, increasing the volume of the Lungs. This causes the pressure inside the Lungs to be less than the outside pressure so air rushes in.

# Physics Chapter Properties Of Matter Answers

**Clifford Lane**



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subject Handpicked numerical are included to upgrade student s knowledge Question bank is given at the end of each chapter I hope that this book will be found usefull by **Prentice Hall Physical Science Concepts in Action Program Planner National Chemistry Physics Earth Science** ,2003-11 Prentice Hall Physical Science Concepts in Action helps students make the important connection between the science they read and what they experience every day Relevant content lively explorations and a wealth of hands on activities take students understanding of science beyond the page and into the world around them Now includes even more technology tools and activities to support differentiated instruction Polymer Science: A Comprehensive Reference ,2012-12-05 The progress in polymer science is revealed in the chapters of Polymer Science A Comprehensive Reference Ten Volume Set In Volume 1 this is reflected in the improved understanding of the properties of polymers in solution in bulk and in confined situations such as in thin films Volume 2 addresses new characterization techniques such as high resolution optical microscopy scanning probe microscopy and other procedures for surface and interface characterization Volume 3 presents the great progress achieved in precise synthetic polymerization techniques for vinyl monomers to control macromolecular architecture the development of metallocene and post metallocene catalysis for olefin polymerization new ionic polymerization procedures and atom transfer radical polymerization nitroxide mediated polymerization and reversible addition fragmentation chain transfer systems as the most often used controlled living radical polymerization methods Volume 4 is devoted to kinetics mechanisms and applications of ring opening polymerization of heterocyclic monomers and cycloolefins ROMP as well as to various less common polymerization techniques Polycondensation and non chain polymerizations including dendrimer synthesis and various click procedures are covered in Volume 5 Volume 6 focuses on several aspects of controlled macromolecular architectures and soft nano objects including hybrids and bioconjugates Many of the achievements would have not been possible without new characterization techniques like AFM that allowed direct imaging of single molecules and nano objects with a precision available only recently An entirely new aspect in polymer science is based on the combination of bottom up methods such as polymer synthesis and molecularly programmed self assembly with top down structuring such as lithography and surface templating as presented in Volume 7 It encompasses polymer and nanoparticle assembly in bulk and under confined conditions or influenced by an external field including thin films inorganic organic hybrids or nanofibers Volume 8 expands these concepts focusing on applications in advanced technologies e g in electronic industry and centers on combination with top down approach and functional properties like conductivity Another type of functionality that is of rapidly increasing importance in polymer science is introduced in volume 9 It deals with various aspects of polymers in biology and medicine including the response of living cells and tissue to the contact with biofunctional particles and surfaces The last volume is devoted to the scope and potential provided by environmentally benign and green polymers as well as energy related polymers They discuss new technologies needed for a sustainable economy in our world of limited resources Provides broad and in depth coverage of all

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*Mechanics and Properties of Matter* Frank Tyler,1961

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**The Physical Properties of Colloidal Solutions** Eli Franklin Burton,1921

**Engineering Digest** ,1964

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