

Molarity Practice Worksheet

Find the molarity of the following solutions:

$$\frac{\text{mol solute}}{\text{L solution}}$$

- 1) 0.5 moles of sodium chloride is dissolved to make 0.05 liters of solution.

$$\frac{0.5 \text{ mol NaCl}}{0.05 \text{ L}} = 10 \text{ M} = 10 \frac{\text{mol}}{\text{L}}$$

- 2) 0.5 grams of sodium chloride is dissolved to make 0.05 liters of solution.

$$\frac{0.5 \text{ g NaCl}}{0.05 \text{ L}} \times \frac{1 \text{ mol NaCl}}{58.44 \text{ g NaCl}} = 0.17 \text{ M}$$

- 3) 0.5 grams of sodium chloride is dissolved to make 0.05 mL of solution.

$$\frac{0.5 \text{ g}}{58.44 \text{ g}} \times \frac{1 \text{ mol}}{1} = \frac{0.0085 \text{ mol}}{0.00005} = 170 \text{ M}$$

$0.05 \text{ mL} \times \frac{1 \text{ L}}{10^3 \text{ mL}}$

Molarity Molality Practice Problems Answers

Scott C. Dulebohn



Molarity Molality Practice Problems Answers:

The Enigmatic Realm of **Molarity Molality Practice Problems Answers**: Unleashing the Language is Inner Magic

In a fast-paced digital era where connections and knowledge intertwine, the enigmatic realm of language reveals its inherent magic. Its capacity to stir emotions, ignite contemplation, and catalyze profound transformations is nothing lacking extraordinary. Within the captivating pages of **Molarity Molality Practice Problems Answers** a literary masterpiece penned with a renowned author, readers set about a transformative journey, unlocking the secrets and untapped potential embedded within each word. In this evaluation, we shall explore the book's core themes, assess its distinct writing style, and delve into its lasting affect the hearts and minds of people who partake in its reading experience.

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