

# 13.2 THE NATURE OF LIQUIDS

## Section Review

### Objectives

- Identify factors that determine physical properties of a liquid
- Define *evaporation* in terms of kinetic energy
- Describe the equilibrium between a liquid and its vapor
- Identify the conditions under which boiling occurs

### Vocabulary

- vaporization
- evaporation
- vapor pressure
- boiling point
- normal boiling point

### Part A Completion

Use this completion exercise to check your understanding of the concepts and terms that are introduced in this section. Each blank can be completed with a term, short phrase, or number.

Liquids are much 1 than gases. Liquids and solids are known as 2 states of matter. The conversion of a liquid to a gas or vapor is called 3. When a liquid that is not 4 changes to a gas, the process is called evaporation. A liquid evaporates faster when heated; however, evaporation itself is a 5 process. When a partially filled container of liquid is sealed, some of the particles at the 6 of the liquid vaporize. These particles collide with the walls of the container, producing a force called 7. The vapor pressure of a liquid can be determined by a device called a 8. A liquid boils when its 9 equals the external pressure. The normal boiling point of a liquid is the temperature at which the vapor pressure is equal to 10.

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10. \_\_\_\_\_

### Part B True-False

Classify each of these statements as always true, AT; sometimes true, ST; or never true, NT.

- \_\_\_\_\_ 11. Particles in a liquid don't have enough kinetic energy to overcome the attractive forces between them and vaporize.

# Nature Of Liquids Section Review Key

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**Ebook: Chemistry: The Molecular Nature of Matter and Change** Silberberg,2015-01-16 Ebook Chemistry The Molecular Nature of Matter and Change *The Vitreous State* Faraday Society,1970 **Federal Register** ,1987-08-19 Practical Druggist and Pharmaceutical Review of Reviews ,1915 **Practical Druggist and Pharmaceutical Review of Reviews** Benjamin Lillard,1919 **Fossil Energy Update** ,1985 **Medical Review of Reviews** ,1925 Index medicus in v 1 30 1895 1924 The Changing Earth: Teacher's ed ,2005 **Digest; Review of Reviews Incorporating Literary Digest** ,1897 *Nature* Sir Norman Lockyer,1904 *Aeronautical Engineering Review* ,1957 Strongly Interacting Quantum Systems out of Equilibrium Thierry Giamarchi,Andrew J. Millis,Olivier Parcollet,Hubert Saleur,Leticia F. Cugliandolo,2016-07-07 Over the last decade new experimental tools and theoretical concepts are providing new insights into collective nonequilibrium behavior of quantum systems The exquisite control provided by laser trapping and cooling techniques allows us to observe the behavior of condensed bose and degenerate Fermi gases under nonequilibrium drive or after quenches in which a Hamiltonian parameter is suddenly or slowly changed On the solid state front high intensity short time pulses and fast femtosecond probes allow solids to be put into highly excited states and probed before relaxation and dissipation occur Experimental developments are matched by progress in theoretical techniques ranging from exact solutions of strongly interacting nonequilibrium models to new approaches to nonequilibrium numerics The summer school Strongly interacting quantum systems out of equilibrium held at the Les Houches School of Physics as its XCIX session was designed to summarize this progress lay out the open questions and define directions for future work This books collects the lecture notes of the main courses given in this summer school *Quarterly Review of the Methodist Episcopal Church, South* ,1884 **The Quarterly Review of the Methodist Episcopal Church, South** ,1884 *Petroleum Review* ,1913 **American Monthly Review of Reviews** ,1902

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