

Stoichiometry

Chem 1019 Section 1

Section 10: Introduction to Stoichiometry

In the space provided, write the letter of the best response that best completes each problem or best answers each question.

1. The number of moles of H_2 that will react with 1 mole of O_2 to form 2 moles of H_2O is:
- a. 1 mole
 - b. 2 moles
 - c. 4 moles
 - d. 8 moles
2. The number of moles of H_2 that will react with 1 mole of O_2 to form 2 moles of H_2O is:
- a. 1 mole
 - b. 2 moles
 - c. 4 moles
 - d. 8 moles
3. The number of moles of H_2 that will react with 1 mole of O_2 to form 2 moles of H_2O is:
- a. 1 mole
 - b. 2 moles
 - c. 4 moles
 - d. 8 moles
4. The number of moles of H_2 that will react with 1 mole of O_2 to form 2 moles of H_2O is:
- a. 1 mole
 - b. 2 moles
 - c. 4 moles
 - d. 8 moles
5. The number of moles of H_2 that will react with 1 mole of O_2 to form 2 moles of H_2O is:
- a. 1 mole
 - b. 2 moles
 - c. 4 moles
 - d. 8 moles
6. The number of moles of H_2 that will react with 1 mole of O_2 to form 2 moles of H_2O is:
- a. 1 mole
 - b. 2 moles
 - c. 4 moles
 - d. 8 moles
7. The number of moles of H_2 that will react with 1 mole of O_2 to form 2 moles of H_2O is:
- a. 1 mole
 - b. 2 moles
 - c. 4 moles
 - d. 8 moles
8. The number of moles of H_2 that will react with 1 mole of O_2 to form 2 moles of H_2O is:
- a. 1 mole
 - b. 2 moles
 - c. 4 moles
 - d. 8 moles
9. The number of moles of H_2 that will react with 1 mole of O_2 to form 2 moles of H_2O is:
- a. 1 mole
 - b. 2 moles
 - c. 4 moles
 - d. 8 moles
10. The number of moles of H_2 that will react with 1 mole of O_2 to form 2 moles of H_2O is:
- a. 1 mole
 - b. 2 moles
 - c. 4 moles
 - d. 8 moles



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