



**Problem Set 4**  
**Chapters 4 and 25**  
**Due Thursday Oct 24<sup>th</sup>**

**Nuclear Chemistry**  
**Essay Due Oct 30<sup>th</sup>**  
**Test Wednesday October 30<sup>th</sup>**

- Your work is to be done on separate sheets of paper.
- All of the papers should be stapled together in the upper left hand corner with this question sheet at the back of the packet.
- Each of your responses should be numbered the same way as it appears on this packet.

**Conceptual Questions**

- 1) State the law of conservation of mass.
- 2) Why are atoms unstable?
- 3) How do unstable atoms gain stability?
- 4) A radioactive element having atomic number 82 and atomic mass 214 loses a beta particle,  ${}_{0}^{-1}\beta$ . What is the resulting particle?
- 5) What type of decay is the reaction below, and what isotope completes this nuclear equation?  
 ${}_{27}^{13}\text{Al} + 2\text{H} \rightarrow 4\text{He} + ?$

**Practice Questions**

- 6) Complete this chart on the three types of radioactive decay

Type	Symbol	Mass (amu)	Charge
Alpha			
Beta			
Gamma			

- 7) If the radioactive atom  ${}_{92}^{238}\text{U}$  emits an alpha particle, the atom remaining is represented by  
(A)  ${}_{92}^{238}\text{U}$  (B)  ${}_{92}^{236}\text{Th}$  (C)  ${}_{92}^{234}\text{U}$  (D)  ${}_{90}^{234}\text{Th}$  (E)  ${}_{91}^{234}\text{Pa}$
- 8) What term completes this equation?  ${}_{14}^{30}\text{Si} + ? \rightarrow {}_{16}^{32}\text{S} + {}_{0}^{-1}e$   
(A)  $4\text{He}$  (B)  ${}^2\text{H}$  (C)  ${}^1\text{H}$  (D)  ${}_{0}^{-1}e$
- 9) What particle is emitted when lead-214,  ${}_{82}^{214}\text{Pb}$ , undergoes natural radioactive decay to bismuth-214,  ${}_{83}^{214}\text{Bi}$ ?  
(A) an alpha particle,  ${}^4\text{He}$  (C) a proton,  ${}^1\text{H}$   
(B) a beta particle,  ${}_{0}^{-1}e$  (D) a neutron,  ${}^1_0\text{n}$

**Extension Question**

- 10) Recently Alexander Valterovich Litvinenko was poisoned by radioactive polonium-210.
  - a. Research who he was and why he was poisoned.
  - b. Find out what polonium-210 does to people.

# Chapter 25 Nuclear Chemistry Worksheet

**Richard J. Kowalsky, J. Randolph Perry**

## Chapter 25 Nuclear Chemistry Worksheet:

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