

Chapter 6 Practice Test: Chemical Bonding

In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question.

- _____ 1. The charge on an ion is
a. always positive.
b. always negative.
c. either positive or negative.
d. zero.
- _____ 2. According to the octet rule, a calcium atom has a tendency to
a. lose one electron.
b. lose two electrons.
c. gain one electron.
d. gain two electrons.
- _____ 3. If a compound forms by ionic bonding, which is *not* true?
a. A positively charged atom or group of atoms attracts a negatively charged atom or group of atoms.
b. The net charge of the compound is zero.
c. The compound contains just two atoms, each of opposite charge.
d. Several ions group together in a tightly packed structure.
- _____ 4. The only property listed that is *not* characteristic of ionic compounds is
a. high melting point.
b. hardness.
c. lack of crystal structure.
d. good conductor of electricity.
- _____ 5. Which formula listed below represents a polyatomic ion?
a. NH_4^+
b. H_2SO_4
c. NaCl
d. H_2O
- _____ 6. The melting points of ionic compounds are higher than the melting points of molecular compounds because
a. ionic substances tend to vaporize at room temperature.
b. ionic substances are brittle.
c. attractive forces between ions are greater than the attractive forces between molecules.
d. the numbers of positive and negative charges are equal in an ionic compound.

Chapter 6 Chemical Bonds Answers

Rex M. Heyworth



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web flappy bird 2012 yılında dong nguyen tarafından geliştirilen popüler bir ios ve android arcade oyunudur basit oynanış ve dinamizm dünya çapında milyonlarca oyuncuyu kendine çekiyor amaç faby kuşunu yeşil boru sıraları arasında yönlendirmek [play flappy bird official online game submit your highscore](#) - Jul 18 2022

web flappy bird is an online game where you simply click or tap the screen to flap your bird through pipes once you ve managed to get through as many pipes as possible be sure to submit your highscore to the leaderboards