

PERIODIC TABLE  **STRAND: UNDERSTANDING PERIODICITY** **ACTIVITY**

Section 1.1 (continued)

In your notebook, read aloud the modern periodic table.

Copy the letterheads in the boxes on the left below from the periodic table to complete the table on the right.

atomic mass	_____
atomic number	_____
electron configuration	_____
chemical name	_____
chemical symbol	_____

Put each item in Column A, under the letter of the matching item in Column B.

Column A

Column B

- | | |
|---|----------------------------------|
| _____ 16. A column in the periodic table | _____ a. metals |
| _____ 17. A row in the periodic table | _____ b. group |
| _____ 18. Group 1 elements | _____ c. period |
| _____ 19. Elements that are shiny and conduct electricity | _____ d. representative elements |
| _____ 20. Group 8 elements | _____ e. transition elements |

In the space on the left, write each of the statements in terms of the statement in bold. Change the underlined word or phrase to make it true.

- _____ 21. There are two representative elements of platinum.
- _____ 22. Most elements border of the elements in the periodic table are metalloids.
- _____ 23. Group 1A elements except for hydrogen are located on the left of metals.
- _____ 24. Group 2A elements are the alkaline earth metals.
- _____ 25. Group 7A elements are highly reactive nonmetals known as halogens.
- _____ 26. Group 8A elements are (noble) unreactive elements located on reactive metals.
- _____ 27. Metalloids have properties of both metals and nonmetals.

Chapter 6 Study Guide For Content Mastery Answers

**Kenneth Raymond Miller, Prentice Hall
(School Division)**

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Ate Wh Holt Rinehart & Winston, 2000 **Glencoe Science**, 2002 Im/Tb Cult Anthropology Gary Ferraro, 2005-04

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Table of Contents Chapter 6 Study Guide For Content Mastery Answers

1. Understanding the eBook Chapter 6 Study Guide For Content Mastery Answers
 - The Rise of Digital Reading Chapter 6 Study Guide For Content Mastery Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Chapter 6 Study Guide For Content Mastery Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chapter 6 Study Guide For Content Mastery Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chapter 6 Study Guide For Content Mastery Answers
 - Personalized Recommendations
 - Chapter 6 Study Guide For Content Mastery Answers User Reviews and Ratings
 - Chapter 6 Study Guide For Content Mastery Answers and Bestseller Lists
5. Accessing Chapter 6 Study Guide For Content Mastery Answers Free and Paid eBooks
 - Chapter 6 Study Guide For Content Mastery Answers Public Domain eBooks
 - Chapter 6 Study Guide For Content Mastery Answers eBook Subscription Services
 - Chapter 6 Study Guide For Content Mastery Answers Budget-Friendly Options
6. Navigating Chapter 6 Study Guide For Content Mastery Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Chapter 6 Study Guide For Content Mastery Answers Compatibility with Devices
 - Chapter 6 Study Guide For Content Mastery Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chapter 6 Study Guide For Content Mastery Answers
 - Highlighting and Note-Taking Chapter 6 Study Guide For Content Mastery Answers
 - Interactive Elements Chapter 6 Study Guide For Content Mastery Answers

8. Staying Engaged with Chapter 6 Study Guide For Content Mastery Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chapter 6 Study Guide For Content Mastery Answers
9. Balancing eBooks and Physical Books Chapter 6 Study Guide For Content Mastery Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chapter 6 Study Guide For Content Mastery Answers
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Chapter 6 Study Guide For Content Mastery Answers
 - Setting Reading Goals Chapter 6 Study Guide For Content Mastery Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Chapter 6 Study Guide For Content Mastery Answers
 - Fact-Checking eBook Content of Chapter 6 Study Guide For Content Mastery Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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