

Chemical Equations and Reactions

Section 2

General Knowledge

Answer the following questions for the space provided.

1. Match the formulae on the left with an appropriate description on the right.

- | | |
|-----------------------|--|
| _____ Ca | (A) a gas formed |
| _____ H_2 | (B) a precipitate formed |
| _____ CO_2 | (C) a solid precipitate formed |
| _____ CaCO_3 | (D) a white solid formed during a reaction |
| _____ CaO | (E) a solid formed in a reaction |

2. From the following list, select the following compound:

_____ MgSO_4 _____ Al^{3+} _____ Mg^{2+} _____ Fe^{2+}

3. In the light of the following reactions, write the usual formula of potassium:

- _____ KNO_3
- _____ K_2O
- _____ K_2SO_4
- _____ K_2CO_3

4. Balance the following equations:



Chapter 8 Review Chemical Equations Answer

Aftab S. Hassan, James L. Flowers



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