

Atomic Structure & Periodic Table Worksheet

1. The table below shows the structure of different atoms and ions. Using the periodic table, complete the table with missing values.

| Element | Atomic Number | Proton Number | Neutron Number | Electron Number | Mass Number |
|------------------|---------------|---------------|----------------|-----------------|-------------|
| Mg | 12 | | 12 | | |
| Mg ²⁺ | 12 | | | | 24 |
| F | | | 10 | | |
| F ⁻¹ | | | 10 | | 19 |

2. The table below gives the composition of three different particles X, Y, and Z.

| Particle | Proton Number | Neutron Number | Electron Number |
|----------|---------------|----------------|-----------------|
| X | 15 | 16 | 15 |
| Y | 15 | 16 | 18 |
| Z | 15 | 17 | 15 |

State the evidence in the table for each of the following statements.

- a) Particle X is an atom.

- b) Particle Y is a negative ion.

- c) X, Y, and Z are all particles of the same element.

- d) Particles X and Z are isotopes of the same elements.

- e) Particles X and Z have zero charges.

Chapter Test Atom And Periodic Table

David Baud



Chapter Test Atom And Periodic Table :

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