



## The Chemistry of Life

## Chapter Test A

### Multiple Choice

Directions: Select the best answer for each question by circling the number of the best possible answer.

- \_\_\_\_\_ 1. If an atom has a mass number of 23 and 11 electrons, the atom is electrically \_\_\_\_\_.
- neutral
  - positive
  - negative
  - ionic
- \_\_\_\_\_ 2. If an atom has a mass number of 23 and 11 protons, the atom is electrically \_\_\_\_\_.
- neutral
  - positive
  - negative
  - ionic
- \_\_\_\_\_ 3. Which of the following is not a constituent of water?
- hydrogen
  - oxygen
  - carbon
  - hydrogen peroxide
- \_\_\_\_\_ 4. A molecule is said to be neutral if the molecule \_\_\_\_\_.
- has no net charge
  - has a net positive charge
  - has a net negative charge
  - has a net ionic charge
- \_\_\_\_\_ 5. Which molecule is not polar, with two \_\_\_\_\_.
- oxygen atoms being slightly positive and two hydrogen atoms being slightly negative
  - oxygen and hydrogen atoms being slightly positive
  - oxygen and hydrogen atoms being slightly negative
  - oxygen and being slightly negative and two hydrogen atoms being slightly positive
- \_\_\_\_\_ 6. If you are not sure which answer is \_\_\_\_\_.
- positive rather than negative
  - negative rather than positive
  - positive rather than neutral
  - neutral rather than positive
- \_\_\_\_\_ 7. A substance with a pH of 7 is called \_\_\_\_\_.
- acidic
  - neutral
  - alkaline
  - amphoteric
- \_\_\_\_\_ 8. Hydrogen chloride when dissolved in water \_\_\_\_\_.
- ionizes
  - forms
  - reacts
  - remains
- \_\_\_\_\_ 9. Why is coffee considered compared to other beverages?
- Coffee does not contain any acids or bases other than different carbohydrates.
  - Coffee does not have quaternary bonds.
  - It is a weak acid because it has a low concentration of hydrogen ions.
  - It is a weak base because it has a low concentration of hydroxide ions.

# Chapter Test Chemistry Of Life Answer Key

**Mike Jess**



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**Chemistry and Life** John William Hill, Dorothy M. Feigl, Stuart J. Baum, 1993      **Chemistry and Life** John W. Hill, 1997

Life William K. Purves, 2001 Authoritative thorough and engaging Life The Science of Biology achieves an optimal

balance of scholarship and teachability never losing sight of either the science or the student The first introductory text to present biological concepts through the research that revealed them Life covers the full range of topics with an integrated experimental focus that flows naturally from the narrative This approach helps to bring the drama of classic and cutting edge research to the classroom but always in the context of reinforcing core ideas and the innovative scientific thinking behind them Students will experience biology not just as a litany of facts or a highlight reel of experiments but as a rich coherent discipline

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## **Table of Contents Chapter Test Chemistry Of Life Answer Key**

1. Understanding the eBook Chapter Test Chemistry Of Life Answer Key
  - The Rise of Digital Reading Chapter Test Chemistry Of Life Answer Key
  - Advantages of eBooks Over Traditional Books
2. Identifying Chapter Test Chemistry Of Life Answer Key
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Chapter Test Chemistry Of Life Answer Key
  - User-Friendly Interface
4. Exploring eBook Recommendations from Chapter Test Chemistry Of Life Answer Key

- Personalized Recommendations
  - Chapter Test Chemistry Of Life Answer Key User Reviews and Ratings
  - Chapter Test Chemistry Of Life Answer Key and Bestseller Lists
5. Accessing Chapter Test Chemistry Of Life Answer Key Free and Paid eBooks
- Chapter Test Chemistry Of Life Answer Key Public Domain eBooks
  - Chapter Test Chemistry Of Life Answer Key eBook Subscription Services
  - Chapter Test Chemistry Of Life Answer Key Budget-Friendly Options
6. Navigating Chapter Test Chemistry Of Life Answer Key eBook Formats
- ePub, PDF, MOBI, and More
  - Chapter Test Chemistry Of Life Answer Key Compatibility with Devices
  - Chapter Test Chemistry Of Life Answer Key Enhanced eBook Features
7. Enhancing Your Reading Experience
- Adjustable Fonts and Text Sizes of Chapter Test Chemistry Of Life Answer Key
  - Highlighting and Note-Taking Chapter Test Chemistry Of Life Answer Key
  - Interactive Elements Chapter Test Chemistry Of Life Answer Key
8. Staying Engaged with Chapter Test Chemistry Of Life Answer Key
- Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Chapter Test Chemistry Of Life Answer Key
9. Balancing eBooks and Physical Books Chapter Test Chemistry Of Life Answer Key
- Benefits of a Digital Library
  - Creating a Diverse Reading Collection Chapter Test Chemistry Of Life Answer Key
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Chapter Test Chemistry Of Life Answer Key
- Setting Reading Goals Chapter Test Chemistry Of Life Answer Key
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Chapter Test Chemistry Of Life Answer Key

- Fact-Checking eBook Content of Chapter Test Chemistry Of Life Answer Key
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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