



**Problem Set 4**  
**Chapters 4 and 25**  
**Due Thursday Oct 24<sup>th</sup>**

**Nuclear Chemistry**  
**Essay Due Oct 30<sup>th</sup>**  
**Test Wednesday October 30<sup>th</sup>**

- Your work is to be done on separate sheets of paper.
- All of the papers should be stapled together in the upper left hand corner with this question sheet at the back of the packet.
- Each of your responses should be numbered the same way as it appears on this packet.

**Conceptual Questions**

- 1) State the law of conservation of mass.
- 2) Why are atoms unstable?
- 3) How do unstable atoms gain stability?
- 4) A radioactive element having atomic number 82 and atomic mass 214 loses a beta particle,  ${}^0_{-1}\beta$ . What is the resulting particle?
- 5) What type of decay is the reaction below, and what isotope completes this nuclear equation?  
 ${}^{27}_{13}\text{Al} + 2\text{H} \rightarrow 4\text{He} + ?$

**Practice Questions**

- 6) Complete this chart on the three types of radioactive decay

Type	Symbol	Mass (amu)	Charge
Alpha			
Beta			
Gamma			

- 7) If the radioactive atom  ${}^{238}_{92}\text{U}$  emits an alpha particle, the atom remaining is represented by  
 (A)  ${}^{238}_{92}\text{U}$     (B)  ${}^{236}_{92}\text{Th}$     (C)  ${}^{234}_{92}\text{U}$     (D)  ${}^{234}_{90}\text{Th}$     (E)  ${}^{234}_{91}\text{Pa}$
- 8) What term completes this equation?  ${}^{30}_{14}\text{Si} + ? \rightarrow 32\text{S} + {}^0_{-1}e$   
 (A)  $4\text{He}$     (B)  $2\text{H}$     (C)  $1\text{H}$     (D)  ${}^0_{-1}e$
- 9) What particle is emitted when lead-214,  ${}^{214}_{82}\text{Pb}$ , undergoes natural radioactive decay to bismuth-214,  ${}^{214}_{83}\text{Bi}$ ?  
 (A) an alpha particle,  ${}^4_2\text{He}$     (C) a proton,  ${}^1_1\text{H}$   
 (B) a beta particle,  ${}^0_{-1}e$     (D) a neutron,  ${}^1_0\text{n}$

**Extension Question**

- 10) Recently Alexander Valterovich Litvinenko was poisoned by radioactive polonium-210.
  - a. Research who he was and why he was poisoned.
  - b. Find out what polonium-210 does to people.

# Chapter 25 Nuclear Chemistry Worksheet

**University of Michigan. College of  
Literature, Science, and the Arts**



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