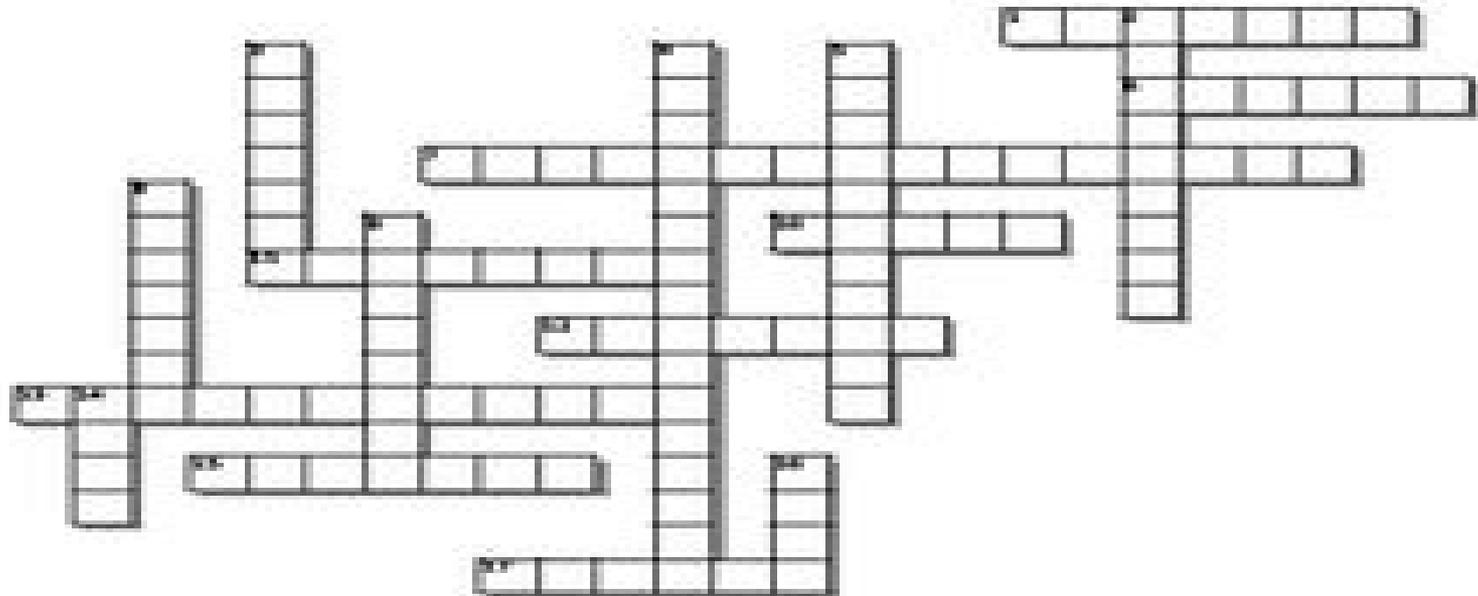


Chapter 2 - The Chemistry of Life (Part 2)



ACROSS

1. Scale with values from 0 to 14, used to measure the concentration of H^+ ions in a solution; a pH of 0 to 7 is acidic, a pH of 7 is neutral, and a pH of 7 to 14 is basic.
6. Compound that prevents sharp, sudden changes in pH.
7. Process that changes, or transforms, one set of chemicals into another set of chemicals.
10. Macromolecule made mostly from carbon and hydrogen atoms; includes fats, oils, and waxes.
11. Elements or compounds that enter into a chemical reaction.
12. Elements or compounds produced by a chemical reaction.
13. Compound made up of carbon, hydrogen, and oxygen atoms; type of nutrient that is the major source of energy for the body (44, 602).
15. Macromolecule that contains carbon, hydrogen, oxygen, and nitrogen; needed

by the body for growth and repair.

17. Protein catalyst that speeds up the rate of specific biological reactions.

DOWN

2. Reactant of an enzyme-catalyzed reaction.
3. Molecules composed of many monomers; makes up macromolecules.
4. Energy that is needed to get a reaction started.
5. Macromolecules containing hydrogen, oxygen, nitrogen, carbon, and phosphorus. Examples include DNA and RNA.
8. Small chemical unit that makes up a polymer.
9. Substance that speeds up the rate of a chemical reaction.
14. Compound that forms hydrogen ions (H^+) in solution; a solution with a pH of less than 7.
16. Compound that produces hydroxide ions (OH^-) in solution; solution with a pH of more than 7.

WORD BANK: ACID, ACTIVATION ENERGY, BASE, BUFFER, CARBOHYDRATE, CATALYST, CHEMICAL REACTION, ENZYME, LIPID, MONOMER, NUCLEIC ACID, PHOSPHATE, POLYMER, PRODUCT, PROTEIN, REACTANT, SUBSTRATE

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