



# COVALENT BONDING

## Practice Problems

Do your best to solve each of the following problems.

### SECTION 8.1 MOLECULAR COMPOUNDS

- Classify each of the following as either ionic or molecular.
  - $\text{NaCl}$
  - $\text{H}_2\text{O}$
  - $\text{HCl}$
  - $\text{KNO}_3$
  - $\text{H}_2\text{SO}_4$
  - $\text{H}_2\text{O}_2$
  - $\text{C}_2\text{H}_6$
  - $\text{CaCl}_2$
  - $\text{H}_2\text{S}$
  - $\text{CO}_2$

- Classify each of the following as either ionic or molecular compound. Write the name of the compound.
- $\text{K}_2\text{SO}_4$
- $\text{C}_2\text{H}_6$
- $\text{H}_2\text{O}$
- $\text{H}_2\text{SO}_4$
- $\text{H}_2\text{O}_2$
- $\text{CaCl}_2$
- $\text{H}_2\text{S}$
- $\text{CO}_2$

### SECTION 8.2 THE NATURE OF COVALENT BONDING

- Show covalent bonding in hydrogen fluoride.  $\text{H}-\text{F}$
- Show covalent bonding in methane.  $\text{CH}_4$
- Show covalent bonding in carbon dioxide.  $\text{CO}_2$
- Show covalent bonding in acetylene.  $\text{H}-\text{C}\equiv\text{C}-\text{H}$
- Draw Lewis structures for each of the following.
  - $\text{H}_2\text{O}$
  - $\text{H}_2\text{O}_2$
  - $\text{C}_2\text{H}_6$
  - $\text{C}_2\text{H}_4$
  - $\text{C}_2\text{H}_2$
  - $\text{C}_2\text{H}_6\text{O}$
  - $\text{C}_2\text{H}_6\text{N}_2$
  - $\text{C}_2\text{H}_4\text{N}_2$
  - $\text{C}_2\text{H}_4\text{O}_2$
  - $\text{C}_2\text{H}_2\text{O}_2$
  - $\text{C}_2\text{H}_2\text{N}_2$
  - $\text{C}_2\text{H}_2\text{O}$
  - $\text{C}_2\text{H}_2\text{N}$
  - $\text{C}_2\text{H}_2\text{O}_2$
  - $\text{C}_2\text{H}_2\text{N}_2$
  - $\text{C}_2\text{H}_2\text{O}$
  - $\text{C}_2\text{H}_2\text{N}$



### SECTION 8.3 BONDING: THE ORBITALS

- Which orbital and sub-orbital are represented by each of the following?
  - $2s$
  - $2p_x$
  - $2p_y$
  - $2p_z$
  - $3s$
  - $3p_x$
  - $3p_y$
  - $3p_z$
  - $4s$
  - $4p_x$
  - $4p_y$
  - $4p_z$
- Which orbital and sub-orbital are represented by each of the following?
  - $3d_{xy}$
  - $3d_{yz}$
  - $3d_{zx}$
  - $3d_{x^2-y^2}$
  - $3d_{z^2}$

# Chapter 8 Covalent Bonding Test A Answers

## Hazwoperore

**Wolfgang Guggemos**



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