

Percent composition, Empirical and Molecular Formula Quiz

Find the percent composition of all of the elements in the following compounds (1 pt per line)



Al $26.982 \times 2 = 53.964 \div 294.15$

S $32.065 \times 3 = 96.195 \div 294.15$

O $15.999 \times 9 = 143.991 \div 294.15$
294.15

Al: 18.35%

S: 32.70%

O: 48.95%



Na $22.99 \times 1 = 22.990 \div 102.$

Br $79.904 \times 1 = 79.904 \div 102.894$
102.894

Na: 22.34%

Br: 77.66%



Fe $55.845 \times 1 = 55.845 \div 115.853$

C $12.011 \times 1 = 12.011 \div 115.853$

O $15.999 \times 3 = 47.997 \div 115.853$
115.853

Fe: 48.20%

C: 10.37%

O: 41.43%



C $12.011 \times 1 = 12.011$

O $15.999 \times 2 = 31.998$
44.009

C: 27.29%

O: 72.71%

Chapter 8 Quiz 2 11 12 Practice Quiz Answer Key

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Quasney**

Chapter 8 Quiz 2 11 12 Practice Quiz Answer Key:

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