

Assignment #1 – Compound Names and Formulas
Single-valent ions only

A. Name these Compounds

- Li_2S Lithium Sulfide
- CaO Calcium Oxide
- NaF Sodium Fluoride
- CaBr_2 Calcium Bromide
- MgCl_2 Magnesium Chloride
- BBr_3 Boron Bromide
- Cs_2O Cesium Oxide
- FrBr Francium Bromide
- Ag_2S Silver Sulfide

- GeF_4 Germanium Fluoride
- Ga_2O_3 Gallium Oxide
- EsCl_3 Einsteinium Chloride
- Fm_2O_3 Fermium Oxide
- Mg_3N_2 Magnesium Nitride
- Rb_2O Rubidium Oxide
- RaO Radium Oxide
- SrO Strontium Oxide
- Tc_2O_7 Technetium Oxide

B. Write the correct chemical formula for these compounds by balancing the ionic charges

- sodium chloride NaCl
- magnesium fluoride MgF_2
- silver oxide Ag_2O
- indium bromide InBr_3
- zinc bromide ZnBr_2
- neodymium oxide Nd_2O_3
- thorium sulphide ThS_2
- actinium oxide Ac_2O_3
- radium bromide RaBr_2
- cesium oxide Cs_2O

- hydrogen oxide H_2O
- francium nitride Fr_3N
- rubidium phosphide Rb_3P
- potassium oxide K_2O
- beryllium sulphide BeS
- lithium sulphide Li_2S
- hydrogen bromide HBr
- strontium nitride Sr_3N_2
- calcium oxide CaO
- tantalum nitride Ta_3N_5

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Table of Contents Chemfax Ionic Formula Writing Kit Answers

1. Understanding the eBook Chemfax Ionic Formula Writing Kit Answers
 - The Rise of Digital Reading Chemfax Ionic Formula Writing Kit Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Chemfax Ionic Formula Writing Kit Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chemfax Ionic Formula Writing Kit Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chemfax Ionic Formula Writing Kit Answers
 - Personalized Recommendations
 - Chemfax Ionic Formula Writing Kit Answers User Reviews and Ratings
 - Chemfax Ionic Formula Writing Kit Answers and Bestseller Lists
5. Accessing Chemfax Ionic Formula Writing Kit Answers Free and Paid eBooks
 - Chemfax Ionic Formula Writing Kit Answers Public Domain eBooks
 - Chemfax Ionic Formula Writing Kit Answers eBook Subscription Services
 - Chemfax Ionic Formula Writing Kit Answers Budget-Friendly Options
6. Navigating Chemfax Ionic Formula Writing Kit Answers eBook Formats

- ePub, PDF, MOBI, and More
 - Chemfax Ionic Formula Writing Kit Answers Compatibility with Devices
 - Chemfax Ionic Formula Writing Kit Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chemfax Ionic Formula Writing Kit Answers
 - Highlighting and Note-Taking Chemfax Ionic Formula Writing Kit Answers
 - Interactive Elements Chemfax Ionic Formula Writing Kit Answers
 8. Staying Engaged with Chemfax Ionic Formula Writing Kit Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chemfax Ionic Formula Writing Kit Answers
 9. Balancing eBooks and Physical Books Chemfax Ionic Formula Writing Kit Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chemfax Ionic Formula Writing Kit Answers
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine Chemfax Ionic Formula Writing Kit Answers
 - Setting Reading Goals Chemfax Ionic Formula Writing Kit Answers
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of Chemfax Ionic Formula Writing Kit Answers
 - Fact-Checking eBook Content of Chemfax Ionic Formula Writing Kit Answers
 - Distinguishing Credible Sources
 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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