

**CHEMFILE MINI-GUIDE TO PROBLEM SOLVING****2. PLAN**

- *What step is needed to calculate the concentration of the diluted solution?*

Apply the principle that  $\text{volume}_1 \times \text{molarity}_1 = \text{volume}_2 \times \text{molarity}_2$ .

$$M_1 V_1 = M_2 V_2 \xrightarrow[\substack{\text{rearrange the equation} \\ \text{algebraically to solve for } M_2 \\ \text{substitute each of the known} \\ \text{quantities for its symbol and} \\ \text{calculate}}]{M_1 V_1 = M_2 V_2} M_2 = \frac{M_1 V_1}{V_2}$$

$$M_2 = \frac{4.74 \text{ M} \times 50.00 \text{ mL}}{250.00 \text{ mL}}$$

**3. COMPUTE**

Note: Even though molarity is moles per liter, you can use volumes in milliliters along with molarity whenever the units cancel.

$$M_2 = \frac{4.74 \text{ M} \times 50.00 \text{ mL}}{250.00 \text{ mL}} = 0.948 \text{ M}$$

**4. EVALUATE**

- *Are the units correct?*
- *Is the number of significant figures correct?*
- *Is the answer reasonable?*

Yes; molarity (mol/L) was required.  
 Yes; the number of significant figures is correct because the data had a minimum of three significant figures.  
 Yes; the computation is the same as  $4.74/5$ , which is a little less than 1.

**PRACTICE**

1. Complete the table below by calculating the missing value in each row.

	<b>Molarity of stock solution</b>	<b>Volume of stock solution</b>	<b>Molarity of dilute solution</b>	<b>Volume of dilute solution</b>	
a.	0.500 M KBr	20.00 mL	7 M KBr	100.00 mL	ans: 0.100 M
b.	1.00 M LiOH	? mL	0.075 M LiOH	500.00 mL	ans: 38 mL
c.	7 M HI	5.00 mL	0.0493 M HI	100.00 mL	ans: 0.986 M
d.	12.0 M HCl	0.250 L	1.8 M HCl	? L	ans: 1.7 L
e.	7.44 M NH <sub>3</sub>	? mL	0.093 M NH <sub>3</sub>	4.00 L	ans: 50. mL

# Chemfile Mini Guide To Problem Solving Answer Key

**Sandra Yancy McGuire**



## **Chemfile Mini Guide To Problem Solving Answer Key:**

**Holt Chemistry File**, 1998 This reference is a must for students who need extra help reteaching or extra practice The guide moves students through the same concepts as the text but at a slower pace More descriptive detail along with visual algorithms provides a more structured approach Each chapter closes with a large bank of practice problems Book jacket

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web i know there is quadratic time solution for some versions of string to string correction problem but with these modifications i m 100 sure its in npc its even listed as npc

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in computer science the string to string correction problem refers to determining the minimum cost sequence of edit operations necessary to change one string into another i e computing the shortest edit distance each type of edit operation has its own cost value a single edit operation may be changing a single symbol of the string into another cost wc deleting a symbol cost wd or inserting a new symbol cost wi

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web jun 29 1990 this leads to the notion of a cyclic string and in this paper we present an  $O(nm \log m)$  algorithm to solve the string to string correction problem for cyclic

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web the string to string correction problem is to determine the distance between two strings as measured by the minimum cost sequence of edit operations needed to change the

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web may 19 2019 i was having the same problem and i found out why you should use double when you are creating `tiptotal` double is not the same as `double` since double is a

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web feb 25 2010 the problem is `arrteammembers` is an array of strings while `tb text` is simply a string you need to assign `tb text` to an index in the array to do this use the

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web in this paper we study a variant of the string sorting problem where deletions and single character interchanges a special case of adjacent block transpositions are the allowed

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web an algorithm is presented which solves the string to string correction problem in time proportional to the product of the lengths of the two strings the string to string

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