

# STOICHIOMETRY: MIXED PROBLEMS

Name \_\_\_\_\_



What volume of  $\text{NH}_3$  at STP is produced if 25.0 g of  $\text{N}_2$  is reacted with an excess of  $\text{H}_2$ ?

\_\_\_\_\_



If 5.0 g of  $\text{KClO}_3$  is decomposed, what volume of  $\text{O}_2$  is produced at STP?

\_\_\_\_\_

3. How many grams of  $\text{KCl}$  are produced in Problem 2?

\_\_\_\_\_



What volume of hydrogen at STP is produced when 2.5 g of zinc react with an excess of hydrochloric acid?

\_\_\_\_\_



How many molecules of water are produced if 2.0 g of sodium sulfate are produced in the above reaction?

\_\_\_\_\_



If 10.0 g of aluminum chloride are decomposed, how many molecules of  $\text{Cl}_2$  are produced?

\_\_\_\_\_

# Chemistry If8766 Stoichiometry Answers

**David Callaghan**



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