

Chemical Analysis – What can you Remember?

1. What colour precipitate do chloride ions produce?
 - a) White
 - b) Cream
 - c) Yellow
 - d) Green
2. What colour do sodium ions go in a flame test?
 - a) Red
 - b) Yellow
 - c) Brick-red
 - d) Lilac
3. Which two solutions are added to test for the presence of halide ions?
 - a) Silver chloride and nitric acid
 - b) Barium chloride and hydrochloric acid
 - c) Silver nitrate and nitric acid
 - d) Barium chloride and sulphuric acid
4. What is the test for oxygen gas?
 - a) Lit splint goes out with a squeaky pop
 - b) Damp blue litmus paper goes red
 - c) Damp red litmus paper goes blue
 - d) Glowing splint relights
5. What colour precipitate do iodide ions form?
 - a) White
 - b) Cream
 - c) Yellow
 - d) Green
6. Which ion produces a brick-red colour in a flame test?
 - a) Sodium
 - b) Calcium
 - c) Copper
 - d) Lithium
7. What is the test for carbon dioxide gas?
 - a) Limewater turns cloudy
 - b) Damp blue litmus turns red
 - c) A glowing splint relights
 - d) A lit splint goes out with a squeaky pop
8. When sodium hydroxide is added to a solution containing iron (III) ions, what colour precipitate forms?
 - a) White
 - b) Blue
 - c) Green
 - d) Red/brown
9. What is the test for chlorine gas?
 - a) Damp blue litmus paper turns red and is then bleached white
 - b) Red litmus turns blue
 - c) A glowing splint relights
 - d) Limewater will turn cloudy
10. Copper (II) ions form what colour precipitate test?
 - a) Blue
 - b) Green
 - c) Red/brown
 - d) Cream
11. Which two solutions are added to test for the presence of sulphate ions?
 - a) Silver chloride and nitric acid
 - b) Barium chloride and sulphuric acid
 - c) Silver nitrate and nitric acid
 - d) Barium chloride and hydrochloric acid
12. What colour do potassium ions go in a flame test?
 - a) Lilac
 - b) Red
 - c) Yellow
 - d) Brick-red
13. What is the test for hydrogen gas?
 - a) Lit splint goes out with a squeaky pop
 - b) Glowing splint relights
 - c) Damp blue litmus paper turns red and then is bleached white
 - d) Limewater goes cloudy
14. Which ion produces a red colour in a flame test?
 - a) Sodium
 - b) Calcium
 - c) Copper
 - d) Lithium
15. What colour precipitate do bromide ions form?
 - a) White
 - b) Cream
 - c) Yellow
 - d) Green
16. What needs to be added to test for the presence of carbonate ions?
 - a) Silver nitrate
 - b) Universal indicator
 - c) Litmus paper
 - d) Dilute acid
17. If sodium hydroxide is added to a compound containing ammonium ions, what gas is produced?
 - a) Oxygen
 - b) Ammonia
 - c) Carbon dioxide
 - d) Hydrogen
18. What colour precipitate does barium sulphate form?
 - a) White
 - b) Green
 - c) Yellow
 - d) Cream
19. What is needed to carry out a flame test?
 - a) Splint
 - b) Glass rod
 - c) Spatula
 - d) Nichrome/platinum wire

Your score out of 19=
Percentage =

Chemistry Solutions Quiz

L Darling-Hammond



Chemistry Solutions Quiz:

The book delves into Chemistry Solutions Quiz. Chemistry Solutions Quiz is a vital topic that must be grasped by everyone, ranging from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Chemistry Solutions Quiz, encompassing both the fundamentals and more intricate discussions.

1. The book is structured into several chapters, namely:

- Chapter 1: Introduction to Chemistry Solutions Quiz
- Chapter 2: Essential Elements of Chemistry Solutions Quiz
- Chapter 3: Chemistry Solutions Quiz in Everyday Life
- Chapter 4: Chemistry Solutions Quiz in Specific Contexts
- Chapter 5: Conclusion

2. In chapter 1, this book will provide an overview of Chemistry Solutions Quiz. This chapter will explore what Chemistry Solutions Quiz is, why Chemistry Solutions Quiz is vital, and how to effectively learn about Chemistry Solutions Quiz.
3. In chapter 2, the author will delve into the foundational concepts of Chemistry Solutions Quiz. This chapter will elucidate the essential principles that must be understood to grasp Chemistry Solutions Quiz in its entirety.
4. In chapter 3, the author will examine the practical applications of Chemistry Solutions Quiz in daily life. This chapter will showcase real-world examples of how Chemistry Solutions Quiz can be effectively utilized in everyday scenarios.
5. In chapter 4, the author will scrutinize the relevance of Chemistry Solutions Quiz in specific contexts. This chapter will explore how Chemistry Solutions Quiz is applied in specialized fields, such as education, business, and technology.
6. In chapter 5, this book will draw a conclusion about Chemistry Solutions Quiz. This chapter will summarize the key points that have been discussed throughout the book.

This book is crafted in an easy-to-understand language and is complemented by engaging illustrations. This book is highly recommended for anyone seeking to gain a comprehensive understanding of Chemistry Solutions Quiz.

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