

Use your notebook to complete all of the following review questions.  
Please note that your completed notebook, corrected quizzes and unit review sheets are the **BEST** study tools. These questions will only help to reinforce some of the exam topics.

1. Draw a line continuum to show the metric prefixes in order from largest (on the left) to smallest (on the right).



2. Complete the following metric conversions:

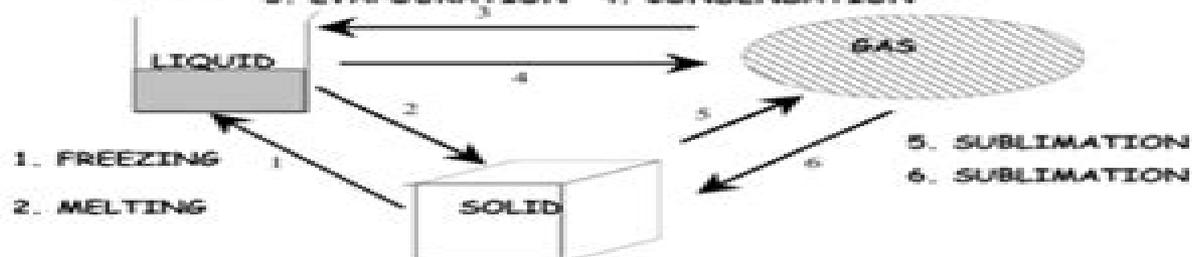
- a) 75 g = 0.075 kg      e) 4.2 kL = 420 000 cL  
 b) 4530 cm = 0.0453 km      f) 83.2 cm = 832 mm  
 c) 0.8 A = 800 mA      g) 975 mg = 0.975 g  
 d) 3200 mA = 3.2 A      h) 975 mg = 0.000975 kg

3. Complete the following table using your knowledge of the Particle Theory of Matter:

| STATE of MATTER | FORCES OF ATTRACTION | SPACING OF MOLECULES  | MOVEMENT OF MOLECULES |
|-----------------|----------------------|-----------------------|-----------------------|
| solid           | <b>strongest</b>     | <b>close together</b> | <b>least</b>          |
| liquid          | <b>weaker</b>        | <b>farther apart</b>  | <b>more</b>           |
| gas             | <b>weakest</b>       | <b>farthest apart</b> | <b>most</b>           |

4. Changes of state are examples of physical changes.

- a) Complete the following diagram with labels and arrows to show all of the SIX changes of state. 3. **EVAPORATION** 4. **CONDENSATION**



5. Name FIVE ways in which you know a chemical change has probably taken place.

colour change      light given off  
bubbles form      heat produced  
precipitate forms      \_\_\_\_\_

# Chemistry Unit Test Grade 9 Answers

**E Durkheim**



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## **Table of Contents Chemistry Unit Test Grade 9 Answers**

1. Understanding the eBook Chemistry Unit Test Grade 9 Answers
  - The Rise of Digital Reading Chemistry Unit Test Grade 9 Answers
  - Advantages of eBooks Over Traditional Books
2. Identifying Chemistry Unit Test Grade 9 Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Chemistry Unit Test Grade 9 Answers
  - User-Friendly Interface
4. Exploring eBook Recommendations from Chemistry Unit Test Grade 9 Answers
  - Personalized Recommendations
  - Chemistry Unit Test Grade 9 Answers User Reviews and Ratings
  - Chemistry Unit Test Grade 9 Answers and Bestseller Lists

5. Accessing Chemistry Unit Test Grade 9 Answers Free and Paid eBooks
  - Chemistry Unit Test Grade 9 Answers Public Domain eBooks
  - Chemistry Unit Test Grade 9 Answers eBook Subscription Services
  - Chemistry Unit Test Grade 9 Answers Budget-Friendly Options
6. Navigating Chemistry Unit Test Grade 9 Answers eBook Formats
  - ePub, PDF, MOBI, and More
  - Chemistry Unit Test Grade 9 Answers Compatibility with Devices
  - Chemistry Unit Test Grade 9 Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Chemistry Unit Test Grade 9 Answers
  - Highlighting and Note-Taking Chemistry Unit Test Grade 9 Answers
  - Interactive Elements Chemistry Unit Test Grade 9 Answers
8. Staying Engaged with Chemistry Unit Test Grade 9 Answers
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Chemistry Unit Test Grade 9 Answers
9. Balancing eBooks and Physical Books Chemistry Unit Test Grade 9 Answers
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Chemistry Unit Test Grade 9 Answers
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Chemistry Unit Test Grade 9 Answers
  - Setting Reading Goals Chemistry Unit Test Grade 9 Answers
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Chemistry Unit Test Grade 9 Answers
  - Fact-Checking eBook Content of Chemistry Unit Test Grade 9 Answers
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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