

Chemistry
1st Semester Practice Exam

- In the following list, only _____ is not an example of matter.
 - planets
 - light
 - dust
 - elemental phosphorus
 - table salt
- What is the physical state in which matter has no specific shape but does have a specific volume?
 - gas
 - solid
 - liquid
 - salts
 - ice
- A combination of sand, salt, and water is an example of a _____.
 - homogeneous mixture
 - heterogeneous mixture
 - compound
 - pure substance
 - solid
- Which one of the following is a pure substance?
 - concrete
 - wood
 - salt water
 - elemental copper
 - milk
- Which one of the following is often easily separated into its components by simple techniques such as filtering or decanting?
 - heterogeneous mixture
 - compounds
 - homogeneous mixture
 - elements
 - solutions
- An element cannot _____.
 - be part of a heterogeneous mixture
 - be part of a homogeneous mixture
 - be separated into other substances by chemical means
 - interact with other elements to form compounds
 - be a pure substance
- In the following list, only _____ is not an example of a chemical reaction.
 - dissolution of a penny in nitric acid
 - the condensation of water vapor
 - a burning candle
 - the formation of polyethylene from ethylene
 - the explosive reaction of hydrogen with oxygen, which produces water,
- Which one of the following is not a physical property of water?
 - It boils at 100°C at 1 atm pressure.
 - It freezes at 0°C at 1 atm pressure.
 - It is clear and colorless.
 - Water exists in solid, liquid and gaseous forms.
 - It reacts rapidly with potassium metal to form potassium hydroxide.
- Which of the following is a physical property of sodium chloride?
 - It is a solid at room temperature.
 - It dissolves in water.
 - It melts at a high temperature.
 - It is not significantly compressible.
 - All of the above are physical properties of sodium chloride.
- Of the following, only _____ is a chemical reaction.
 - melting of lead
 - dissolving sugar in water
 - tarnishing of silver
 - crushing of stone
 - dropping a penny into a glass of water

Chemistry 1st Semester Practice Exam

John Moore, Richard H. Langley



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