

Chapter 3 – Scientific Measurement

Chapter 3: 1 – 24, 26 - 28, 32, 34, 38, 40, 42, 46, 51, 56, 57, 62, 85, 87 (39 total)

Section Review 3.1

1. a. What is the difference between a *qualitative* measurement and a *quantitative* measurement?

Qualitative measurements are expressed in descriptive, non-numerical form, whereas quantitative measurements are expressed in a definite form, typically numerical.

- b. How is a number converted to scientific notation?

It is written as the product of a coefficient greater than or equal to one and less than ten, plus ten raised to a power.

2. Classify each statement as either qualitative or quantitative.

- a. The basketball is brown. *qualitative*
 b. The diameter of the basketball is 31 centimeters. *quantitative*
 c. The air pressure in the basketball is 12 pounds per square inch. *quantitative*
 d. The surface of the basketball has indented seams. *qualitative*

3. Write each measurement in scientific notation.

- a. the length of a football field, 91.4 meters. 9.14×10^1 meters
 b. the diameter of a carbon atom, 0.000 000 000 154 meter. 1.54×10^{-10} meters
 c. the radius of Earth, 6 378 000 meters 6.378×10^6 meters
 d. the diameter of a human hair, 0.000 008 meter. 8×10^{-6} meters
 e. the average distance between the center of the sun and the center of Earth, 149 600 000 000 meters. 1.496×10^{11} meters

4. Solve each problem, and express each answer in correct scientific notation.

- a. $(4 \times 10^7)(2 \times 10^{-3}) = 8 \times 10^4$
 b. $(6.3 \times 10^{-2}) / 2.1 \times 10^4 = 3.0 \times 10^{-6}$
 c. $(4.6 \times 10^3) - (1.8 \times 10^3) = 2.8 \times 10^3$
 d. $(7.1 \times 10^{-2}) + (5 \times 10^{-3}) = 7.6 \times 10^{-2}$

Chemistry Chapter 3 Scientific Measurement Test

Clemens Wendtner



Chemistry Chapter 3 Scientific Measurement Test:

Decoding **Chemistry Chapter 3 Scientific Measurement Test**: Revealing the Captivating Potential of Verbal Expression

In an era characterized by interconnectedness and an insatiable thirst for knowledge, the captivating potential of verbal expression has emerged as a formidable force. Its power to evoke sentiments, stimulate introspection, and incite profound transformations is genuinely awe-inspiring. Within the pages of "**Chemistry Chapter 3 Scientific Measurement Test**," a mesmerizing literary creation penned by a celebrated wordsmith, readers attempt an enlightening odyssey, unraveling the intricate significance of language and its enduring effect on our lives. In this appraisal, we shall explore the book's central themes, evaluate its distinctive writing style, and gauge its pervasive influence on the hearts and minds of its readership.

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