

Chapter 7: Solutions and Colloids

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CHAPTER OUTLINE

7.1 Physical States of Solutions	7.4 Solution Concentrations	7.7 Solution Properties
7.2 Solubility	7.5 Solution Preparation	7.8 Colloids
7.3 The Solution Process	7.6 Solution Stoichiometry	7.9 Dialysis

LEARNING OBJECTIVES/ASSESSMENT

When you have completed your study of this chapter, you should be able to:

1. Classify mixtures as solutions or nonsolutions based on their appearance. (Section 7.1; Exercise 7.4)
2. Demonstrate your understanding of terms related to the solubility of solutes in solution. (Section 7.2; Exercises 7.6 and 7.12)
3. Predict in a general way the solubilities of solutes in solvents on the basis of molecular polarity. (Section 7.3; Exercise 7.16)
4. Calculate solution concentrations in units of molarity, weight/weight percent, weight/volume percent, and volume/volume percent. (Section 7.4; Exercises 7.22 b, 7.30 c, 7.34 a, and 7.38 c)
5. Describe how to prepare solutions of specific concentration using pure solutes and solvent, or solutions of greater concentration than the one desired. (Section 7.5; Exercises 7.46 and 7.48 b)
6. Do stoichiometric calculations based on solution concentrations. (Section 7.6; Exercise 7.56)

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