

## 1. Basic Notation of Light

- 1. Light is the transfer of energy.
- 2. The main properties of light include frequency and wavelength. Frequency is generally denoted the wavelength by the Greek letter  $\lambda$  or  $\lambda_{\text{light}}$  where  $\lambda$  is wavelength,  $\nu$  is frequency, and  $c$  is speed of light  $3 \times 10^8$  m/s.
- 3. Light has two properties: electromagnetic waves and particles. It is called light in collective terms.
- 4. The wavelength is denoted by the Greek letter  $\lambda$  or  $\lambda_{\text{light}}$  where  $\lambda$  is wavelength of the photon. In Planck's constant  $h = 6.626 \times 10^{-34}$  J s, and  $\nu$  is frequency.
- 5. The quantum properties of accelerated particles, such as photons, are described by  $E = h\nu$  where  $E$  is energy,  $h$  is Planck's constant, and  $\nu$  is frequency or wavelength.
- 6. Energy

$E = h\nu$  where  $\nu$  is frequency or  $\lambda$

In Planck's constant  $h = 6.626 \times 10^{-34}$  J s, and  $\nu$  is frequency.

$\lambda$  is wavelength,  $c$  is speed of light  $3 \times 10^8$  m/s

and  $\nu$  is frequency.

$E = h\nu$  where  $E$  is energy

$\lambda$  is wavelength or  $\nu$  is frequency

### Common Questions

1. How does the wavelength of light change as frequency changes? Does the energy of photons also change? Explain.
2. A photon's energy is directly proportional to its frequency. If the frequency of the light is 100 Hz, what is the energy of the photon? Explain.
3. The wavelength of light is directly proportional to its frequency. If the frequency of the light is 100 Hz, what is the wavelength of the light? Explain.
4. A photon's energy is directly proportional to its frequency. If the frequency of the light is 100 Hz, what is the energy of the photon? Explain.
5. In what way is the energy of a photon related to its frequency? Explain.
6. Calculate the energy of a photon with a wavelength of 100 nm. Explain.
7. Calculate the energy of a photon with a wavelength of 100 nm. Explain.
8. Calculate the energy of a photon with a wavelength of 100 nm. Explain.
9. Calculate the energy of a photon with a wavelength of 100 nm. Explain.
10. Calculate the energy of a photon with a wavelength of 100 nm. Explain.
11. Calculate the energy of a photon with a wavelength of 100 nm. Explain.
12. Calculate the energy of a photon with a wavelength of 100 nm. Explain.

# Chemquest 11 Answers

**N Noddings**



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