

Chapter 6: Thermochemistry

Homework: 6.4, 6.10, 6.26, 6.43, 6.52, 6.53; 6.55

1. Define energy and work. (Slide 2 and 4)

2. Describe the difference between kinetic and potential energy. Show by dimensional analysis that they both have the same units.

3. What is the first law of thermodynamics? Use the diagram on slide 12 to explain.

4. List examples of some forms of energy.

5. Homework 6.4: Convert 880 K to your year into (a) J, (b) kJ, (c) Cal (Slide 11)

6. A state function is a mathematical function whose result depends only on the _____ and _____ conditions, not on the process used.

e.g.

7. Energy diagrams: (Slide 18-19)

8. Heat and Work: (Slide 20-22)

9. Define the terms in the equation: q , n and ΔT .

10. Homework 6.45: How much heat is required to warm 1.50 kg of sand from 25.0 °C to 100.0 °C?

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Jessica J Manson



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