

Section 12.1

Section 12.1

Answer complete answer multiple-choice questions that are provided.

1. _____
 - a. _____
 - (a) correct, in terms of all the reactants and products.
 - (b) correct, in terms of moles of reactants and products.
 - (c) correct, in terms of work done by work done by the system.
 - (d) correct, in terms of enthalpy change, ΔH .
 - b. _____
 - (a) the heat capacity of the system.
 - (b) the mass of the system.
 - (c) the number of moles of hydrogen that will react with a certain quantity of oxygen.
 - (d) the initial enthalpy of the system.
 - c. _____
 - (a) correct, in terms of the energy of the system.
 - (b) correct, in terms of the energy of the system.
 - (c) correct, in terms of the energy of the system.
 - (d) correct, in terms of the energy of the system.
 - d. _____
 - (a) correct, in terms of the energy of the system.
 - (b) correct, in terms of the energy of the system.
 - (c) correct, in terms of the energy of the system.
 - (d) correct, in terms of the energy of the system.

2. Answer the two questions based on the following balanced equation:



a. Balance the equation:



b. What is the coefficient of H_2 in the balanced equation?

c. How many moles of H_2 are required to produce 1.0 mole of H_2O ?

d. How many moles of O_2 are required to produce 1.0 mole of H_2O ?

e. How many moles of H_2 are required to produce 1.0 mole of H_2O ?

Chapter 9 Stoichiometry Section 1 Answers

Rachel Sandford



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