



The Chemistry of Life

Chapter Test A

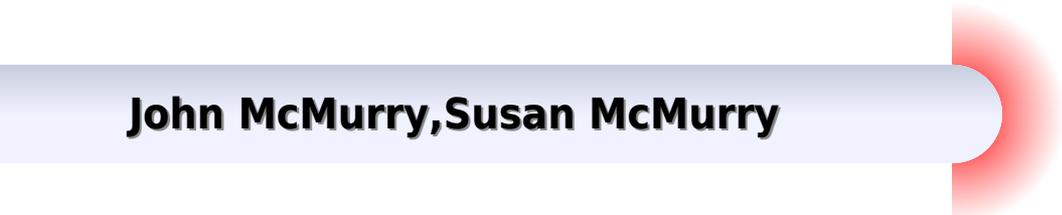
Multiple Choice

Directions: Select the best answer for each of the questions by circling the number of the best possible answer.

1. If an atom has a mass number of 23 and 11 electrons, the atom is most likely to
 - a. be an ion
 - b. be a cation
 - c. be a neutral atom
 - d. be an anion
2. If an atom has a mass number of 23 and 11 protons, the atom is most likely to
 - a. be an ion
 - b. be a cation
 - c. be a neutral atom
 - d. be an anion
3. Which of the following is not a characteristic of water?
 - a. low density of hydrogen-bonded water molecules
 - b. very strong surface tension and surface adhesion
 - c. moderate solubility and high capacity of hydrogen
 - d. moderate solubility and low density of water
4. A molecule that is neutral in the overall
 - a. ionizing solution
 - b. ionizing process
 - c. sharing electrons
 - d. sharing protons
5. Which molecule can polarize water?
 - a. oxygen with being slightly positive and the hydrogen with being slightly negative
 - b. oxygen with hydrogen with being slightly positive
 - c. oxygen with hydrogen with being slightly negative
 - d. oxygen with being slightly negative and the hydrogen with being slightly positive
6. If you are with water, which water are you in?
 - a. positive solute-solvent
 - b. positive and negative
 - c. solvent-solute
 - d. solvent only
7. A substance with a pH of 7 is called
 - a. an acid
 - b. basic
 - c. saline
 - d. a compound
8. Sodium chloride is a(n) _____ that is _____
 - a. molecule
 - b. atom
 - c. salt
 - d. mixture
9. Why is coffee-sugar considered a pure substance?
 - a. Coffee-sugar has both the pure quality and the uniformity of different substances.
 - b. Coffee-sugar has both quality and quantity.
 - c. Both coffee-sugar and fruit-juice-sugar are mixtures of hydrogen and hydrogen.
 - d. Both coffee-sugar and the substance in water-soluble and insoluble.

Chapter Test Chemistry Of Life Answer Key

John McMurry, Susan McMurry



Chapter Test Chemistry Of Life Answer Key:

Chapter Resource 2 Chemistry of Life Biology Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2004

Chemistry and Life John William Hill, Dorothy M. Feigl, Stuart J. Baum, 1993 Chemistry and Life John W. Hill, 1997

Life William K. Purves, 2001 Authoritative thorough and engaging Life The Science of Biology achieves an optimal balance of scholarship and teachability never losing sight of either the science or the student The first introductory text to present biological concepts through the research that revealed them Life covers the full range of topics with an integrated experimental focus that flows naturally from the narrative This approach helps to bring the drama of classic and cutting edge research to the classroom but always in the context of reinforcing core ideas and the innovative scientific thinking behind them Students will experience biology not just as a litany of facts or a highlight reel of experiments but as a rich coherent discipline

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