

Revision Notes Class 11 Chemistry

Chapter 1 - Some Basic Concepts of Chemistry.

1. CHEMISTRY

Chemistry (derived from the Egyptian word kēme (chem), which means "earth") is a science that studies the composition, structure and properties of matter and the changes it undergoes during chemical reactions. Chemistry is often referred to as core science because it plays a role in linking physical sciences (including chemistry) with life sciences and applied sciences (such as medicine and engineering).

Chemistry is divided into following branches:

1.1 Physical chemistry

The branch of chemistry which deals with macroscopic as well as physical phenomena in a universe. It is generally the impact of physical property on the chemical property as well as structure of a substance.

1.2 Inorganic chemistry

The branch of chemistry that studies compounds that do not contain carbon and hydrogen atoms is called "inorganic chemistry." Simply put, it is the opposite of organic chemistry. Substances that do not have carbon-hydrogen bonds include metals, salts, and chemicals.

1.3 Organic chemistry

The discipline which deals with the study of the structure, composition and the chemical properties of organic compounds is known as organic chemistry. It involves the study of Carbon and its compounds.

1.4 Biochemistry

Biochemistry is that branch of chemistry that explores the chemical processes in organisms and associated with them. It's a laboratory-based science that connects biology and chemistry. By using chemical knowledge and technology, biochemists can understand and solve biological problems.

1.5 Analytical chemistry

It is the branch of chemistry which uses instruments and analytical techniques to determine structure, functionality and properties of a substance.

Chemistry Chapter 1 Notes

David Kirk



Chemistry Chapter 1 Notes:

UPSC NCERT Books - Class 8 Notes and Summary Mocktime Publication,2023-01-22 UPSC NCERT Books Class 8 Notes and Summary Notes on books Longmans, Green and co,1903 **Elemental** Tim James,2019-03-26 If you want to understand how our world works the periodic table holds the answers When the seventh row of the periodic table of elements was completed in June 2016 with the addition of four final elements nihonium moscovium tennessine and oganesson we at last could identify all the ingredients necessary to construct our world In Elemental chemist and science educator Tim James provides an informative entertaining and quirkily illustrated guide to the table that shows clearly how this abstract and seemingly jumbled graphic is relevant to our day to day lives James tells the story of the periodic table from its ancient Greek roots when you could count the number of elements humans were aware of on one hand to the modern alchemists of the twentieth and twenty first centuries who have used nuclear chemistry and physics to generate new elements and complete the periodic table In addition to this he answers questions such as What is the chemical symbol for a human What would happen if all of the elements were mixed together Which liquid can teleport through walls Why is the medieval dream of transmuting lead into gold now a reality Whether you re studying the periodic table for the first time or are simply interested in the fundamental building blocks of the universe from the core of the sun to the networks in your brain Elemental is the perfect guide Technology Quarterly and Proceedings of the Society of Arts ,1906 Vol 8 14 include Review of American chemical research edited by Arthur A Noyes **The Technology Quarterly and Proceedings of the Society of Arts** ,1906 **Sea Grant Publications Index, 1968-72** ,1973 **The Journal of Education** ,1905 *Educational Times and Journal of the College of Preceptors* ,1908 **University of California Union Catalog of Monographs Cataloged by the Nine Campuses from 1963 Through 1967: Authors & titles** University of California (System). Institute of Library Research,University of California, Berkeley,1972 **Educational Times** ,1897 **The Bookseller** ,1915 **Plant Science Catalog** National Agricultural Library (U.S.),1958 Quantitative Chemical Analysis Daniel C. Harris,2010-04-30 QCA is the bestselling textbook of choice for analytical chemistry It offers a modern portrait of the techniques of chemical analysis backed by a wealth of real world applications This edition features new coverage of spectroscopy and statistics new pedagogy and enhanced lecturer support **The High School Journal** ,1922 Scientific American ,1913

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findings This helpful guide is for both theoretical chemists and experimental photochemistry researchers interested in utilizing computational photochemistry methods for their own work Reviews the fundamentals of photochemistry helping those new to the field in understanding key concepts Provides detailed guidance and comparison of computational and theoretical methods highlighting the suitability of each method for different case studies Outlines current applications to encourage discussion of the synergy between experimental and computational data and inspiring further application of these methods to other photochemical processes

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