

## Chemistry Worksheets Class 10 on Chapter 1 Chemical Reactions and Equations- Set 1

**Q-1:** Write the balanced chemical equations for each of the following reactions.

- Liquid hydrogen peroxide decomposes to form water and oxygen gas.
- Acetylene gas burns in oxygen to form carbon dioxide and water along with evolution of heat.
- When solid mercury(II) oxide is heated, liquid mercury and oxygen gas are produced.
- Marble dissolves in a hydrochloric acid to give calcium chloride, water and carbon dioxide.
- Copper sulphate on treatment with potassium iodide precipitates cuprous iodide, liberates iodine gas and also forms potassium sulphate.

**Q-2:** Fill in the blanks

- \_\_\_\_\_ is the magnetic oxide of iron.
- In reactivity series, manganese lies \_\_\_\_\_ zinc
- To the reaction between  $\text{AgNO}_3$  solution and  $\text{NaCl}$  solution, the names given to this type of reactions are \_\_\_\_\_ and \_\_\_\_\_.
- Proteins decompose to form \_\_\_\_\_ in our body.
- If copper turnings are added to colourless  $\text{AgNO}_3$  solution, it turns \_\_\_\_\_ after sometime.

**Q-3:** Which of the following displacement reactions will not take place?

- $\text{Zn(s)} + \text{FeSO}_4(\text{aq}) \rightarrow \text{ZnSO}_4(\text{aq}) + \text{Fe(s)}$
- $\text{Fe(s)} + \text{CuSO}_4(\text{aq}) \rightarrow \text{FeSO}_4(\text{aq}) + \text{Cu(s)}$
- $\text{Cu(s)} + 2\text{AgNO}_3(\text{aq}) \rightarrow \text{Cu(NO}_3)_2(\text{aq}) + 2\text{Ag(s)}$
- $\text{Cu(s)} + \text{FeSO}_4(\text{aq}) \rightarrow \text{CuSO}_4(\text{aq}) + \text{Fe(s)}$

**Q-4:** Which of the following statements is false?

- Oxidation results in the loss of an electropositive element and the gain of an electronegative element.
- Reduction results in the loss of an electronegative element and the gain of an electropositive element.
- A species that gains electrons is referred to as a reducing agent.
- An oxidising agent is a type of specie that absorbs hydrogen atoms.

**Q-5:** Why are certain chemicals and medications kept in coloured bottles?

**Q-6:** Sakshi desired that her home be white washed. She went to the market and bought 20 kg of quick lime, which she dissolved in 60 litres of water. When she added lime to water, she noticed that the water began to boil even though it was not heated. Give an explanation for her observation. Name the product and write the corresponding chemical equation and word equation.

# Chemistry Chapter 1 Worksheet

**Huangqi Zhang**



## **Chemistry Chapter 1 Worksheet:**

**Chemistry (Teacher Guide)** Dr. Dennis Englin, 2018-02-26 This book was created to help teachers as they instruct students through the Master's Class Chemistry course by Master Books The teacher is one who guides students through the subject matter helps each student stay on schedule and be organized and is their source of accountability along the way With that in mind this guide provides additional help through the laboratory exercises as well as lessons quizzes and examinations that are provided along with the answers The lessons in this study emphasize working through procedures and problem solving by learning patterns The vocabulary is kept at the essential level Practice exercises are given with their answers so that the patterns can be used in problem solving These lessons and laboratory exercises are the result of over 30 years of teaching home school high school students and then working with them as they proceed through college Guided labs are provided to enhance instruction of weekly lessons There are many principles and truths given to us in Scripture by the God that created the universe and all of the laws by which it functions It is important to see the hand of God and His principles and wisdom as it plays out in chemistry This course integrates what God has told us in the context of this study Features Each suggested weekly schedule has five easy to manage lessons that combine reading and worksheets Worksheets quizzes and tests are perforated and three hole punched materials are easy to tear out hand out grade and store Adjust the schedule and materials needed to best work within your educational program Space is given for assignments dates There is flexibility in scheduling Adapt the days to your school schedule Workflow Students will read the pages in their book and then complete each section of the teacher guide They should be encouraged to complete as many of the activities and projects as possible as well Tests are given at regular intervals with space to record each grade About the Author DR DENNIS ENGLIN earned his bachelor's from Westmont College his master of science from California State University and his EdD from the University of Southern California He enjoys teaching animal biology vertebrate biology wildlife biology organismic biology and astronomy at The Master's University His professional memberships include the Creation Research Society the American Fisheries Association Southern California Academy of Sciences Yellowstone Association and Au Sable Institute of Environmental Studies

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**Prentice Hall Science Explorer** Michael J. Padilla,2002 *Engineering with Mathcad* Brent Maxfield,2006-11-18 Using the author s considerable experience of applying Mathcad to engineering problems Engineering with Mathcad identifies the most powerful functions and features of the software and teaches how to apply these to create comprehensive engineering calculations Many examples from a variety of engineering fields demonstrate the power and utility of Mathcad s tools while also demonstrating how other software such as Microsoft Excel spreadsheets can be incorporated effectively This simple step by step approach makes this book an ideal Mathcad text for professional engineers as well as engineering and science students A CD ROM packaged with the book contains all the

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**Physical Chemistry Calculations** Rodney J. Sime, 2005 Physical Chemistry Calculations is a practical guide for students and instructors who want to learn how to use the most popular spreadsheet and computational software to solve problems in physical chemistry The book provides students with a complementary approach to the chemistry and physics they are learning in the classroom Physical Chemistry Calculations also gives a solid introduction to calculations with Excel VB VBA MathCad and Mathematica

**Modern Chemistry** Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2001

*Radiologic Science* Stewart C. Bushong, 2001-04 This companion to Bushong's RADIOLOGIC SCIENCE FOR TECHNOLOGISTS textbook features the same comprehensiveness as the text The first of three main sections consists of worksheets organized by textbook chapter that allow students to work through the main topics of radiologic science Suitable as either homework or an in class assignment these worksheets can be completed with reference to the text if needed

*Spreadsheet Chemistry* O. Jerry Parker, Gary L. Breneman, 1991

**Spreadsheets for Chemists** Gordon Filby, 1995 A practical guide Spreadsheets for Chemists shows chemists of all levels how to use spreadsheet programs in their daily work It highlights the possibilities provided by Lotus 1 2 3 the most widely used spreadsheet program in the sciences Apart from hundreds of example fragments it features Detailed discussion of the most relevant functions and all the macro commands An accompanying diskette containing 57 worksheets involving many different fields of chemical research and teaching An extensive glossary of spreadsheet terms Three appendices covering 1 2 3 s competitors and add in packages the use of Windows based spreadsheets and how what if analysis and back solving is applied Although the disk examples were developed for Lotus 1 2 3 DOS Versions 2 x x 2 4 the worksheets are compatible with the newer Windows versions and those of their main competitors Borland's Quattro Pro and Microsoft's Excel Several compatible spreadsheets AsEasyAs Proqube might also be as used as inexpensive alternatives The author is a senior scientist at the Nuclear Research Centre in Karlsruhe Germany He has been using spreadsheet software for nearly ten years successfully in a variety of chemical problems

Chapter Resource 2 Chemistry of Life Biology Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2004

*Enrichment Chemistry Modern Course* GLENCOE, 1995-04

**Merrill**

**Chemistry** Robert C. Smoot, Smoot, Richard G. Smith, Jack Price, 1998 First Year Chemistry Students' Conceptions of Acid/base Chemistry Sally Diane Rupert, 2001 **Science Spectrum** Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2003-03 **Hazardous Materials Chemistry** Armando S. Bevelacqua, 2006 This book explores basic principles of chemistry and toxicology so first responders can efficiently identify a chemical hazard anticipate potential outcomes and make safe and informed decisions Back cover **Astronomy** Alton Biggs, 2002

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