

**Chemistry**  
**Chapter 9 Practice Test: Stoichiometry**

**Name** \_\_\_\_\_

**Free Response**

*Use the following equation to answer problems 1 - 4 at STP. (Show All Work)*



1. How many **moles** of aluminum chloride ( $\text{AlCl}_3$ ) would be produced from 0.12 moles of Hydrochloric acid (HCl)?
2. If you wanted to completely react 5.0 moles of Aluminum, how many **moles** of hydrochloric acid (HCl) would be needed?
3. How many **grams** of hydrochloric acid (HCl) would be needed to completely react 35.8 g of Aluminum foil?
4. How many **grams** of aluminum would be needed to produce 67.2 dm<sup>3</sup> of hydrogen gas?

*Use the following equation to answer problems 5 and 6 at STP.*



5. How many **molecules** of carbon dioxide ( $\text{CO}_2$ ) would be produced by completely reacting 8.00 grams of oxygen gas ( $\text{O}_2$ ) with excess Carbon Monoxide (CO)?
6. What **volume** in L of oxygen gas ( $\text{O}_2$ ) would react with 6.0 L of carbon monoxide gas (CO) ?

# Chemistry Chapter 9 Stoichiometry Test

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