

AP Chemistry Practice Test Chapter 8 and 9

Multiple Choice

Identify the choice that best completes the statement or answers the question.

- _____ 1. Which pair of atoms should form the most polar bond?
a. F and B
b. C and O
c. F and O
d. N and F
e. B and N
- _____ 2. Which pair of ions should form the ionic lattice with the highest energy?
a. Na^{1+} and Br^{1-}
b. Li^{1+} and F^{1-}
c. Cs^{1+} and F^{1-}
d. Li^{1+} and O^{2-}
e. K^{1+} and F^{1-}
- _____ 3. The electrostatic force of attraction is greatest in which compound?
a. BaO
b. MgO
c. CaS
d. MgS
e. CaO
- _____ 4. Which molecule has the weakest bond?
a. CO
b. O_2
c. NO
d. N_2
e. Cl_2
- _____ 5. For which species are octet resonance structures necessary and sufficient to describe the bonding satisfactorily?
a. BCl_3
b. SO_2
c. CO_2
d. BeF_2
e. SO_4^{2-}
- _____ 6. How are the bonding pairs arranged in the best Lewis structure for ozone?
a. $\text{O}-\text{O}-\text{O}$
b. $\text{O}=\text{O}-\text{O}$
c. $\text{O}=\text{O}=\text{O}$
d. $\text{O}=\text{O}=\text{O}$
e. $\text{O}=\text{O}=\text{O}$
- _____ 7. Which species has the shortest bond length?
a. CN^{1-}
b. O_2
c. SO_2
d. SO_3
e. CO_2
- _____ 8. Which species requires the least amount of energy to remove an electron from the outermost energy level?
a. Na^{1+}
b. He
c. F^{1-}
d. O^{2-}
e. Mg^{2+}
- _____ 9. Which species has a valid non-octet Lewis structure?
a. GeCl_4
b. SiF_4
c. CO_3^{2-}
d. SeCl_4
e. CO_3^{2-}

Chemistry Chapter 9 Test Answers

Christian Drosten



Chemistry Chapter 9 Test Answers:

Reviewing **Chemistry Chapter 9 Test Answers**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is actually astonishing. Within the pages of "**Chemistry Chapter 9 Test Answers**," an enthralling opus penned by a very acclaimed wordsmith, readers embark on an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve to the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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