

Name Key

1. Hydrogen, H_2 , is used as rocket fuel. The hydrogen is burned in oxygen, O_2 , according to the following equation. What is the enthalpy change if 200.0 g of hydrogen, H_2 is burned?



$$\frac{200.0 \text{ g } H_2}{2.016 \text{ g } H_2} \times \frac{1 \text{ mol } H_2}{1} = 99.2 \text{ mol } H_2$$

$$99.2 \text{ mol } H_2 \times \frac{-484 \text{ kJ}}{2 \text{ mol } H_2} = -24,006 \text{ kJ}$$

$-24,006 \text{ kJ}$
~~24,006 kJ~~

2. Carbon disulfide burns in air, producing carbon dioxide and sulfur dioxide.



What is ΔH for the following reaction?



3. Following is the equation for the combustion of acetone:



How much heat will be given off (in kcal) if 4.5 mol of acetone is burned completely?

$$\frac{-853.6 \text{ kcal}}{2 \text{ mol } C_3H_6O} \times 4.5 \text{ mol } C_3H_6O = -1920.6 \text{ kcal}$$

Exothermic

4. The specific heat of iron is $0.110 \text{ cal} / \text{g} \cdot ^\circ\text{C}$. How much heat must be added to 5.2 g of iron at 25°C to raise its temperature to 225°C ?

$$\text{heat} = q = m c \Delta T = (5.2 \text{ g}) \left(\frac{0.110 \text{ cal}}{\text{g} \cdot ^\circ\text{C}} \right) (225 - 25^\circ\text{C}) = 114.4 \text{ cal}$$

Endothermic

5. Hydrazine, N_2H_4 , is a colorless liquid used as a rocket fuel. What is the enthalpy change for the process in which hydrazine is formed from its elements (reaction (1) below)? Use Hess's Law to calculate ΔH for reaction (1) given the data below.



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Angie Papple Johnston, Rod Powers



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the line (x_1, y_1) and the slope of the line m and want to find other points on the line have a play with it move the point try
different slopes now let s discover more what does it stand for

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web in the points lines and equations gizmo you can plot two points and see the line through them and their coordinates 1
drag a point in the gizmo and describe what happens 2 drag the points to 2 3 and 6 5 what is the equation of this line 3 select
show probe and drag the red probe what do the input x and output y mean

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web a first quadrant coordinate plane the x and y axes each scale by one a graph of a line goes through the points $(0, 3)$, $(3, 0)$, and $(2, 7)$ which are plotted and labeled let s write the equation of the line that passes through the points $(0, 3)$, $(3, 0)$, and $(2, 7)$ in slope intercept form

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web the points we use cartesian coordinates to mark a point on a graph by how far along and how far up it is example the point $(12, 5)$ is 12 units along and 5 units up steps there are 3 steps to find the equation of the straight line 1 find the slope of the line 2 put the slope and one point into the point slope formula

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web sep 27 2020 the given line is written in $mx + b$ the slope is 2 to find the slope of a perpendicular line find the reciprocal $\frac{1}{2}$ the slope of the perpendicular line is $\frac{1}{2}$ use the method for writing an equation from the slope and a point on the line substitute 1 5 for x and y $y = mx + b$ $5 = \frac{1}{2}(1) + b$

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web feb 26 2021 to find the equation of the line pictured in figure page index 2 select an arbitrary point $q(x, y)$ on the line then compute the slope of the line using left (x_1, y_1) right $(p, 0)$ and left (x_2, y_2) right (q, x, y) in the slope formula 1 text slope $\frac{y_2 - y_1}{x_2 - x_1} = \frac{y - b}{x - 0}$ nonnumber

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web find the equation of the line in all three forms listed above two of the forms require slope so let's find that first slope $m = \frac{\Delta y}{\Delta x} = \frac{4 - 2}{9 - 3} = \frac{2}{6} = \frac{1}{3}$ now we can plug in m and one of the points say $(5, 5)$ to get point slope form $y - 5 = \frac{1}{3}(x - 5)$

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web what is the equation for a vertical line the slope is undefined and where does it cross the y axis in fact this is a special case and we use a different equation not y but instead we use x like this $x = 1.5$ every point on the line has x coordinate 1.5 that is why its equation is $x = 1.5$