

Equilibrium

- **Equilibrium** is a state where the net force on an object is zero. It can be static or dynamic.
- **Static Equilibrium**: The object is at rest.
 - $\sum F_x = 0$ and $\sum F_y = 0$
 - $\sum \tau = 0$
- **Dynamic Equilibrium**: The object is moving with constant velocity.
 - $\sum F_x = 0$ and $\sum F_y = 0$
 - $\sum \tau = 0$
- **Center of Mass**: The point where the mass of the object is concentrated.
 - $\sum \tau = 0$ about the center of mass.
- **Stable Equilibrium**: The object returns to its original position after a small displacement.
- **Unstable Equilibrium**: The object moves away from its original position after a small displacement.
- **Neutral Equilibrium**: The object remains in its original position after a small displacement.



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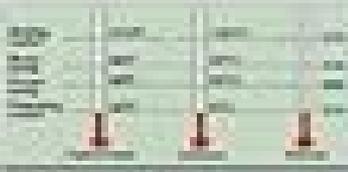
Nuclear Chemistry

- **Radioactivity** is the process by which unstable atomic nuclei lose energy by emitting radiation.
- **Types of Radiation**:
 - **Alpha** (α): ${}^4_2\text{He}$
 - **Beta** (β): ${}^0_{-1}\text{e}$
 - **Gamma** (γ): γ
- **Half-life** ($t_{1/2}$): The time it takes for half of the radioactive substance to decay.
- **Radioactive Decay**:
 - $N = N_0 e^{-\lambda t}$
 - $t_{1/2} = \frac{\ln 2}{\lambda}$
- **Applications**:
 - **Medicine**: Cancer treatment, diagnostic imaging.
 - **Industry**: Radiography, smoke detectors.
 - **Energy**: Nuclear power generation.



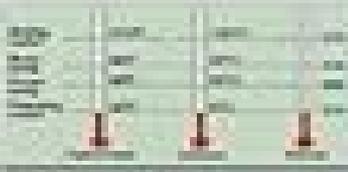
Measurement & Units

- **SI Units**: International System of Units.
- **Base Units**:
 - Length: meter (m)
 - Mass: kilogram (kg)
 - Time: second (s)
 - Temperature: kelvin (K)
 - Amount of substance: mole (mol)
 - Luminous intensity: candela (cd)
- **Derived Units**:
 - Area: square meter (m^2)
 - Volume: cubic meter (m^3)
 - Density: kg/m^3
 - Speed: m/s
 - Acceleration: m/s^2
 - Force: newton (N)
 - Energy: joule (J)
 - Power: watt (W)



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