

Chemical Analysis – What can you Remember?

1. What colour precipitate do chloride ions produce?
a) **White**
b) Cream
c) Yellow
d) Green
2. What colour do sodium ions go in a flame test?
a) Red
b) **Yellow**
c) Brick-red
d) Lilac
3. Which two solutions are added to test for the presence of halide ions?
a) Silver chloride and nitric acid
b) Barium chloride and hydrochloric acid
c) **Silver nitrate and nitric acid**
d) Barium chloride and sulphuric acid
4. What is the test for oxygen gas?
a) Lit splint goes out with a squeaky pop
b) Damp blue litmus paper goes red
c) Damp red litmus paper goes blue
d) **Glowing splint relights**
5. What colour precipitate do iodide ions form?
a) White
b) Cream
c) **Yellow**
d) Green
6. Which ion produces a brick-red colour in a flame test?
a) Sodium
b) **Calcium**
c) Copper
d) Lithium
7. What is the test for carbon dioxide gas?
a) **Lime water turns cloudy**
b) Damp blue litmus turns red
c) A glowing splint relights
d) A lit splint goes out with a squeaky pop
8. When sodium hydroxide is added to a solution containing iron (III) ions, what colour precipitate forms?
a) White
b) Blue
c) Green
d) **Red/brown**
9. What is the test for chlorine gas?
a) **Damp blue litmus paper turns red and is then bleached white**
b) Red litmus \square turns blue
c) A glowing splint relights
d) Lime water will turn cloudy
10. Copper (II) ions form what colour precipitate test?
a) **Blue**
b) Green
c) Red/brown
d) Cream
11. Which two solutions are added to test for the presence of sulphate ions?
a) Silver chloride and nitric acid
b) Barium chloride and sulphuric acid
c) Silver nitrate and nitric acid
d) **Barium chloride and hydrochloric acid**
12. What colour do potassium ions go in a flame test?
a) **Lilac**
b) Red
c) Yellow
d) Brick-red
13. What is the test for hydrogen gas?
a) **Lit splint goes out with a squeaky pop**
b) Glowing splint relights
c) Damp blue litmus paper turns red and then is bleached white
d) Limewater goes cloudy
14. Which ion produces a red colour in a flame test?
a) Sodium
b) Calcium
c) Copper
d) **Lithium**
15. What colour precipitate do bromide ions form?
a) White
b) **Cream**
c) Yellow
d) Green
16. What needs to be added to test for the presence of carbonate ions?
a) Silver nitrate
b) Universal indicator
c) Litmus paper
d) **Dilute acid**
17. If sodium hydroxide is added to a compound containing ammonium ions, what gas is produced?
a) Oxygen
b) **Ammonia**
c) Carbon dioxide
d) Hydrogen
18. What colour precipitate does barium sulphate form?
a) **White**
b) Green
c) Yellow
d) Cream
19. What is needed to carry out a flame test?
a) Splint
b) Glass rod
c) Spatula
d) **Nichrome/platinum wire**

Your score out of 19 =
Percentage =

Chemistry Solutions Quiz

Baylor University



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