

1. Basic Notation of Light

- 1. Light is the transfer of energy.
- 2. The main properties of light include frequency and wavelength. Frequency is generally denoted the wavelength by the Greek letter λ or λ_{light} where λ is wavelength, ν is frequency, and c is speed of light 3×10^8 m/s.
- 3. Light has two properties: wavelength and frequency. It is possible to find a relation between the two.
- 4. The wavelength of a photon is related to frequency of light via the relationship $E = h\nu$ where E is energy of the photon, h is Planck's constant (having 6.626×10^{-34} J s), and ν is frequency.
- 5. The relationship between wavelength and frequency is expressed as $c = \lambda\nu$ where c is speed of light, λ is wavelength, ν is frequency, and c is constant 3×10^8 m/s.
- 6. Energy

$E = h\nu$ where ν is frequency in Hz

$E = hf$ where f is frequency in Hz

$c = \lambda\nu$ where λ is wavelength in m, ν is frequency in Hz

$c = \lambda f$

$E = hf$ where f is

$E = hf$ where f is frequency in Hz

Example Questions

1. The wavelength of a photon is 400 nm. What is the frequency of this photon?
2. A photon has a wavelength of 600 nm. What is the energy of this photon?
3. The wavelength of a photon is 400 nm. What is the energy of this photon?
4. The wavelength of a photon is 400 nm. What is the energy of this photon?
5. A photon has a wavelength of 400 nm. What is the energy of this photon?
6. A photon has a wavelength of 400 nm. What is the energy of this photon?
7. A photon has a wavelength of 400 nm. What is the energy of this photon?
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10. A photon has a wavelength of 400 nm. What is the energy of this photon?
11. A photon has a wavelength of 400 nm. What is the energy of this photon?
12. A photon has a wavelength of 400 nm. What is the energy of this photon?

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Robert R. Redfield



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