

BOOK REVIEWS

Organic Chemistry

Alan S. Wingrove and Robert L. Caret, Harper & Row, New York, NY, 1981, xvii + 1334 pp. Figs. and tables. 18 X 26 cm.

"Organic Chemistry" by Wingrove and Caret enters a most competitive market where it challenges some well-established textbooks widely considered to be among the best examples of the textbook writing art. It is readily apparent that considerable care has gone into the selection and presentation of material by the authors. It is a thoroughly modern, complete text with a large selection of problems, a summary of significant reactions, and supplementary material set apart in red print in each chapter. Red color is also used in illustrations to help the reader better understand the explanations. Although the authors have included an extensive index and thoughtfully chosen supplementary reading lists, the text contains no answers to the problem sets. However, a study guide and answer book is available to aid students in improving and evaluating their study.

Although several techniques are used to divide discussions into manageable portions, the style of presentation and illustration appears less inviting to use than some of the more respected textbooks in current use. The authors have chosen to begin most topics with a theoretical discussion and follow it with specific examples. In my opinion, a strength of the most widely used texts has been their preoccupation with factual information. Theory has been used by these texts to explain and interrelate facts and to enhance understanding rather than serve as the basis for discussing experimental results.

Wingrove and Caret have written a quality text for the first-year course in organic chemistry. The serious student can get a thorough introduction to the field from it. I believe, however, that most students will find this text somewhat more difficult to use than some of its competitors. Nonetheless, I recommend that instructors include this fine text among those given consideration when choosing a new text.

Wilmer K. File

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Experiments in Physical Chemistry, Fourth Edition

David P. Shoemaker, Carl W. Garland,
Jeffrey I. Steinfeld, and Joseph W. Nibler,
McGraw-Hill Book Co., New York, NY,
1981, vii + 787 pp. Figs. and tables. 18.5
X 24.5 cm. \$23.95.

Thomas C. Ehlert
Marquette University
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This book is intended as a textbook for a junior-level laboratory course in physical chemistry. About 80% of the book is devoted to describing fifty tested experiments which span the traditional concerns of physical chemistry, i.e., gases, transport properties of gases, thermochemistry, solutions, phase equilibria, electrochemistry, chemical kinetics, surface phenomena, macromolecules, electric and magnetic properties, spectra and molecular structure, and solids. These clear and well-illustrated descriptions explain in adequate detail the theory, apparatus, procedure, and calculations. They even include details of construction of items not available from supply houses. A list of chemicals and another of readings pertinent to the experiment are also provided.

The balance of the book consists of sections on safety, report writing, treatment of data, electrical measurements, temperature measurements, vacuum techniques, common measuring instruments and procedures, and the use of computers.

Changes from the previous edition, reviewed in THIS JOURNAL, 52, A434 (1975), include the addition of two experiments on NMR spectroscopy, one on laser Raman spectroscopy, and one on single-crystal X-ray crystallography. Four experiments have been dropped, the least squares treatment of data expanded to a full chapter, and the chapter on computers revised and expanded.

The changes from the previous edition are for the better although modest in extent. It is disappointing that the authors did not use the occasion of a new edition to switch to SI units or, better in this transition period, to use SI and non-SI units together. (Factors for conversion to/from SI are given on the inside cover.) And while the chapter on computers is much better than its predecessor it can present only an overview of this subject. Faults in the experiments are hard to find, although for showing the effect of ionic strength on the solubility of a soluble salt the cost and low tendency to dissociate makes silver acetate (authors' choice) inferior to calcium fluoride.

Persons seeking a textbook on this subject will want to examine this book. It may not eliminate the supplements and special directions which seem to be an inevitable part of physical chemistry laboratory, but it will come close.

Reviewed in this Issue

Alan S. Wingrove and Robert L. Caret, Organic Chemistry

David P. Shoemaker, Carl W. Garland, Jeffrey I. Steinfeld and Joseph W. Nibler,
Experiments in Physical Chemistry, Fourth Edition

New Volumes in Continuing Series

Titles of Interest

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Lei Shi



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