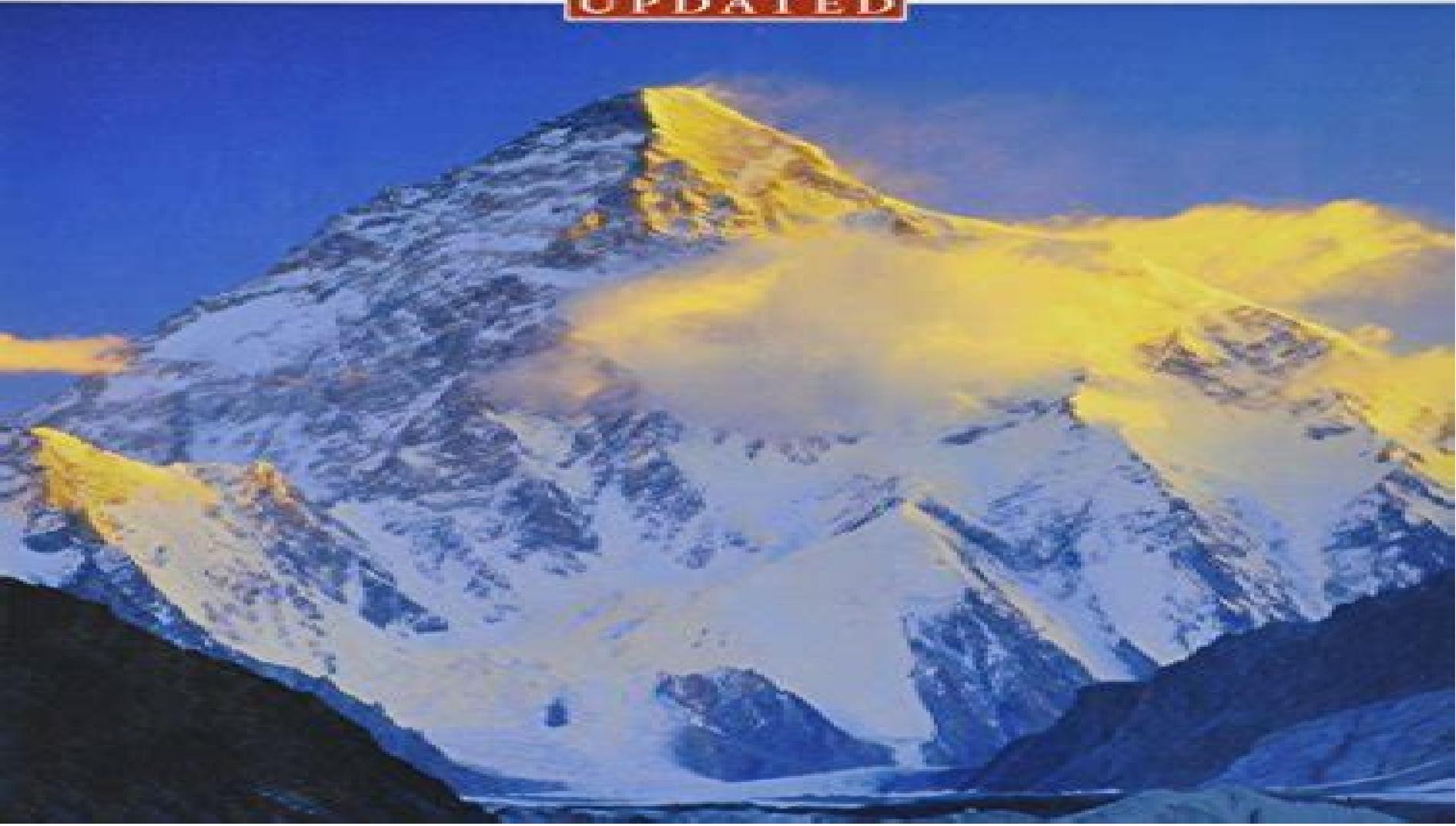


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web balancing redox reactions in acidic solution problems 1 10 problem 1 cr₂o₇²⁻ fe₂ cr₃ fe₃ solution 1 balanced half reactions 6e 14h⁺ cr₂o₇²⁻ 2cr₃ 7h₂o fe₂ fe₃ e 2 equalize the electrons 6e 14h⁺ cr₂o₇²⁻ 2cr₃ 7h₂o 6fe₂ 6fe₃ 6e multiplied by a factor of 6

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web balance each half reaction in basic solution 4 5 6 7 cr2o7 2 no so4 2 mno2 cr3 no3 so2 mn2o3 balance each redox reaction in acid solution using the half reaction method 8 h2o2 cr2o7 2 9 teo3 2 n2o4 o2 cr3 te 10 reo4 11 pbo2 io i2 pb2 12 as no3 io3 re io3 h2aso4

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web practice exercises balance the following equations of redox reactions assign oxidation numbers to all elements in the reaction separate the redox reaction into two half reactions balance the atoms in each half reaction add the two half reactions together and cancel out common terms

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web aug 29 2023 some points to remember when balancing redox reactions the equation is separated into two half equations one for oxidation and one for reduction the equation is balanced by adjusting coefficients and adding h 2 o h and e in this order balance the atoms in the equation apart from o and h

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web in the redox equation below that occurs in our stomach which of the following pairs identifies the reducing and oxidizing agents respectively c 6 h 8 o 6 2h 2no 2 c 6 h 6 o 6 2h 2 o 2no choose 1 answer c 6 h 8 o 6 and no 2 a c 6 h 8 o 6 and no 2 no 2 and c 6 h 8 o 6 b no 2 and c 6 h 8 o 6 h and no 2 c h and