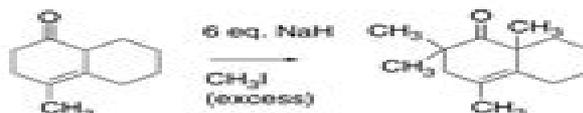


- b. This transformation involves methylation at all enolizable positions. The alkylation was effected using a sixfold excess of NaH and excess methyl iodide. Evidently there is not a significant amount of methylation at C(4), which could occur through  $\gamma$ -alkylation of the C(8a)-enolate.



- c. This alkylation was accomplished using two equivalents of NaNH<sub>2</sub> in liquid NH<sub>3</sub>. The more basic site in the dianion is selectively alkylated. Note that the dianion is an indenyl anion, and this may contribute to its accessibility by di-deprotonation.



- d. This is a nitrile alkylation involving an anion that is somewhat stabilized by conjugation with the indole ring. The anion was formed using NaNH<sub>2</sub> in liquid NH<sub>3</sub>.



1. The starting material is 2-phenyl-1H-inden-5(1H)-one. It reacts with 2 equivalents of NaNH<sub>2</sub> in liquid NH<sub>3</sub> to form a dianion (indenyl anion) with a negative charge on the oxygen atom. This intermediate then reacts with PhCH<sub>2</sub>Cl to form the 2-phenyl-1-(benzylphenyl)inden-5(1H)-one product.
2. The starting material is 2-phenyl-1H-inden-5(1H)-one. It reacts with 2 equivalents of NaNH<sub>2</sub> in liquid NH<sub>3</sub> to form a dianion (indenyl anion) with a negative charge on the oxygen atom. This intermediate then reacts with PhCH<sub>2</sub>Cl to form the 2-phenyl-1-(benzylphenyl)inden-5(1H)-one product.



# Carey Sundberg Solution Manual

**Raffaella Di Napoli**



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## **Table of Contents Carey Sundberg Solution Manual**

1. Understanding the eBook Carey Sundberg Solution Manual
  - The Rise of Digital Reading Carey Sundberg Solution Manual
  - Advantages of eBooks Over Traditional Books
2. Identifying Carey Sundberg Solution Manual
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Carey Sundberg Solution Manual
  - User-Friendly Interface
4. Exploring eBook Recommendations from Carey Sundberg Solution Manual
  - Personalized Recommendations
  - Carey Sundberg Solution Manual User Reviews and Ratings
  - Carey Sundberg Solution Manual and Bestseller Lists
5. Accessing Carey Sundberg Solution Manual Free and Paid eBooks

- Carey Sundberg Solution Manual Public Domain eBooks
  - Carey Sundberg Solution Manual eBook Subscription Services
  - Carey Sundberg Solution Manual Budget-Friendly Options
6. Navigating Carey Sundberg Solution Manual eBook Formats
    - ePub, PDF, MOBI, and More
    - Carey Sundberg Solution Manual Compatibility with Devices
    - Carey Sundberg Solution Manual Enhanced eBook Features
  7. Enhancing Your Reading Experience
    - Adjustable Fonts and Text Sizes of Carey Sundberg Solution Manual
    - Highlighting and Note-Taking Carey Sundberg Solution Manual
    - Interactive Elements Carey Sundberg Solution Manual
  8. Staying Engaged with Carey Sundberg Solution Manual
    - Joining Online Reading Communities
    - Participating in Virtual Book Clubs
    - Following Authors and Publishers Carey Sundberg Solution Manual
  9. Balancing eBooks and Physical Books Carey Sundberg Solution Manual
    - Benefits of a Digital Library
    - Creating a Diverse Reading Collection Carey Sundberg Solution Manual
  10. Overcoming Reading Challenges
    - Dealing with Digital Eye Strain
    - Minimizing Distractions
    - Managing Screen Time
  11. Cultivating a Reading Routine Carey Sundberg Solution Manual
    - Setting Reading Goals Carey Sundberg Solution Manual
    - Carving Out Dedicated Reading Time
  12. Sourcing Reliable Information of Carey Sundberg Solution Manual
    - Fact-Checking eBook Content of Carey Sundberg Solution Manual
    - Distinguishing Credible Sources
  13. Promoting Lifelong Learning
    - Utilizing eBooks for Skill Development

- Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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