

Iodine Clock Reaction

Research question

How does the concentration of sodium bisulfite affect the rate of reaction between acidified potassium iodate solution and sodium bisulfite solution?

Objective

To investigate the effect of the concentration of sodium bisulfite on its rate of reaction with acidified potassium iodate solution, measured by the time it takes for the appearance of blue-black colour in the reaction mixture.

Balanced chemical equation with state symbols



Materials (with uncertainty)

1. Potassium iodate solution (solution A)
2. Sodium bisulfite solution (solution B)
3. Starch solution (solution C)
4. Distilled water
5. 3ml Plastic graduated pipette (± 0.25 ml)
6. Stopwatch (± 0.01 s)
7. Reaction beaker
8. White paper

Volumes of solutions used in the experiment

Experiment	Distilled water (± 0.25 ml)	Solution A/ Potassium iodate solution (± 0.25 ml)	Solution C/ Starch solution (± 0.25 ml)	Solution B/Sodium bisulfite solution (± 0.25 ml)
1	3	1.0	0.5	0.5
2	2.5	1.0	0.5	1
3	2	1.0	0.5	1.5
4	1.5	1.0	0.5	2

Iodine Clock Experiment Lab Report

Michael Hoelscher



Iodine Clock Experiment Lab Report:

Iodine Clock Experiment Lab Report Book Review: Unveiling the Magic of Language

In an electronic era where connections and knowledge reign supreme, the enchanting power of language has been apparent than ever. Its capability to stir emotions, provoke thought, and instigate transformation is truly remarkable. This extraordinary book, aptly titled "**Iodine Clock Experiment Lab Report**," written by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound affect on our existence. Throughout this critique, we shall delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

https://db1.greenfirefarms.com/book/virtual-library/Documents/Pro_Side_Hustles_Guide_For_Students.pdf

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