

## Chapter 4 Atomic Structure

**Section 4.2 The Structure of an Atom****(pages 108–112)**

*This section compares the properties of three subatomic particles. It also discusses atomic numbers, mass numbers, and isotopes.*

**Reading Strategy (page 108)**

**Monitoring Your Understanding** Before you read, list in the table shown what you know about atoms and what you would like to learn. After you read, list what you have learned. For more information on this Reading Strategy, see the **Reading and Study Skills** in the **Skills and Reference Handbook** at the end of your textbook.

What I Know About Atoms	What I Would Like to Learn	What I Have Learned
Most students will know that atoms are the “building blocks” of matter, and some may know that atoms contain subatomic particles.	Based on the title of the section, students may say that they want to learn more about the structure of atoms.	

**Properties of Subatomic Particles (pages 108–109)**

- What are three subatomic particles?  
a. Protons      b. Electrons      c. Neutrons
- Circle the letter that identifies a subatomic particle with a positive charge.  
a. nucleus      **(b)** proton  
c. neutron      d. electron
- Why did Chadwick conclude that the particles produced by his experiment were neutral in charge? A charged object did not deflect the paths of the particles.  
\_\_\_\_\_  
\_\_\_\_\_

**Comparing Subatomic Particles (pages 109–110)**

- Circle the letters of properties that vary among subatomic particles.  
a. color      **(b)** mass  
**(c)** charge      **(d)** location in the atom
- Circle the letter of the expression that accurately compares the masses of neutrons and protons.  
**(a)** mass of 1 neutron = mass of 1 proton  
b. mass of 2000 neutrons = mass of 1 proton  
c. mass of 1 electron = mass of 1 proton  
d. mass of 1 neutron = mass of 1 electron

# Chapter 4 Atomic Structure Answer Key

**Christian G. Meyer**



## **Chapter 4 Atomic Structure Answer Key:**

## Whispering the Strategies of Language: An Emotional Quest through **Chapter 4 Atomic Structure Answer Key**

In a digitally-driven world where screens reign great and quick interaction drowns out the subtleties of language, the profound techniques and emotional nuances concealed within words usually go unheard. However, situated within the pages of **Chapter 4 Atomic Structure Answer Key** a captivating literary value pulsing with fresh thoughts, lies an extraordinary journey waiting to be undertaken. Composed by a skilled wordsmith, this marvelous opus attracts readers on an introspective trip, gently unraveling the veiled truths and profound influence resonating within the material of every word. Within the emotional depths of the emotional evaluation, we shall embark upon a genuine exploration of the book's primary themes, dissect their interesting publishing design, and succumb to the powerful resonance it evokes deep within the recesses of readers' hearts.

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web number of conductor turns in a coil is  $n$  magnetic field strength  $h$   $1 \text{ oe } 103 \text{ 4 a m } 79 \text{ 58 a m mmf magnetic flux } 1 \text{ wb } 1 \text{ vs}$  assignment design a 90 degree bending magnet for beam analysis with the duoplasmatron

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