

## CHAPTER 7 REVIEW

### *Chemical Formulas and Chemical Compounds*

#### MIXED REVIEW

**SHORT ANSWER** Answer the following questions in the space provided.

1. Write formulas for the following compounds:

- \_\_\_\_\_ a. copper(II) carbonate  
 \_\_\_\_\_ b. sodium sulfite  
 \_\_\_\_\_ c. ammonium phosphate  
 \_\_\_\_\_ d. tin(IV) sulfide  
 \_\_\_\_\_ e. nitrous acid

2. Write the Stock names for the following compounds:

- \_\_\_\_\_ a.  $\text{Mg}(\text{ClO}_4)_2$   
 \_\_\_\_\_ b.  $\text{Fe}(\text{NO}_3)_2$   
 \_\_\_\_\_ c.  $\text{Fe}(\text{NO}_2)_3$   
 \_\_\_\_\_ d.  $\text{CoO}$   
 \_\_\_\_\_ e. dinitrogen pentoxide

3. \_\_\_\_\_ a. How many atoms are represented by the formula  $\text{Ca}(\text{HSO}_4)_2$ ?  
 \_\_\_\_\_ b. How many moles of oxygen atoms are in a 0.50 mol sample of this compound?  
 \_\_\_\_\_ c. Assign the oxidation number to sulfur in the  $\text{HSO}_4^-$  anion.

4. Assign the oxidation number to the element specified in each of the following:

- \_\_\_\_\_ a. hydrogen in  $\text{H}_2\text{O}_2$   
 \_\_\_\_\_ b. hydrogen in  $\text{MgH}_2$   
 \_\_\_\_\_ c. sulfur in  $\text{S}_8$   
 \_\_\_\_\_ d. carbon in  $(\text{CO}_3)^{2-}$   
 \_\_\_\_\_ e. chromium in  $\text{Na}_2\text{Cr}_2\text{O}_7$   
 \_\_\_\_\_ f. nitrogen in  $\text{NO}_2$

# Chapter 7 3 Answers Chemical Formulas And Chemical Compounds

**WJ Hussar**



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